# ORDER OF THE STATE OF WISCONSIN NATURAL RESOURCES BOARD REPEALING AND RECREATING, AND CREATING RULES

The Wisconsin Natural Resources Board proposes an order to repeal and recreate NR 219.04 Tables A, B, BM, C, D, E, EM, and F; and to create NR 219.03 (3) and table ES relating to analytical test methods for testing effluent discharges.

# SS-09-04

# Analysis Prepared by the Department of Natural Resources

**Statutory Authority:** Statutes that authorize the promulgation of this rule include ss. 227.11(2)(a), 281.19(1), 283.31, 283.55 (1), and 299.15, Stats. These sections grant rule-making authority to the Department to prevent and abate pollution of the waters of the state. All rules promulgated under this authority are subject to review under ch. 227, Stats.

**Statutes Interpreted:** In promulgating this rule ss. 281.19(1), 283.11(2), 283.55(1)(d), and 299.15(2), Stats., have been interpreted as allowing the Department the authority to specify test methods for testing wastewater for determining contaminate concentration.

**Plain Language Rule Analysis:** Chapter NR 219, Wis. Adm. Code, lists approved test methods to determine the concentration of contaminants in wastewater discharges and sludges. The proposed changes to ch. NR 219 updates the chapter to reflect changes made by the U.S. Environmental Protection Agency to 40 CFR part 136. This proposal adopts revisions to the made in the 04/15/1998, 09/21/1998, 12/30/1999, 01/16/2001, 8/31/2001, 10/23/2002, 10/29/2002, 07/21/2003, and 09/19/2003 Federal Registers. Other EPA methods that have been revised will have their citations updated. In addition, methods for the analysis of PCBs in sludge are included.

The proposed changes to ch. NR 219 potentially affects all certified and registered laboratories and wastewater facilities that collect samples for compliance with their Wisconsin Pollutant Discharge Elimination System (WPDES) permit. There are about 450 certified or registered laboratories. Most of these laboratories solely analyze wastewater samples. There are 276 municipal wastewater, 69 industrial environmental, 91 commercial environmental, 11 public health, and 9 other laboratories. Only the 9 other labs (hazardous waste labs) will not be affected by this rule update. Citing newer analytical methodologies in ch. NR 219 is an improvement that the laboratory community supports.

Federal Regulatory Analysis: The federal counter part to this rule is 40 CFR 136.

**State Regulatory Analysis:** The States of Illinois, Iowa, Michigan, and Minnesota all have similar rules to ch. NR 219 in order to have wastewater programs delegated to them from the U.S. Environmental Protection Agency under the Clean Water Act. Wastewater discharges in these states are tested for contaminant concentration with test methods that are cited in the appropriate State's laws.

A Summary of Factual Data: There is no factual data associated with this rule. Rather this rule revision is to bring Wisconsin up to date with the current federal rules.

**Regulatory Flexibility Analysis:** This proposal does not address any change in reporting, schedule, or deadline requirements. Sections 299.11 and 283.55, Stats., do not allow for less stringent schedules, deadlines, or reporting requirements. However, there is a provision in s. NR 219.05 to apply for a variance from the requirements of ch. NR 219. Under federal and state rules, approval from the U.S. Environmental Protection Agency is required for a variance.

SECTION 1. NR 219.03 (3) is created to read:

# (3) "Sludge" is defined in ss. NR 204.03 (55) and NR 214.03 (34).

# SECTION 2. NR 219.04 TABLE A is repealed and recreated to read:

Parameter and Units	Method <sup>1</sup>	EPA	Standard Methods 18 <sup>th</sup> , 19 <sup>th</sup> , 20 <sup>th</sup> Ed.	USGS	Other
Bacteria:					
1. Coliform (fecal) number per 100 ml	Most Probable Number (MPN), 5 tube, 3 dilution; or,	p.132 <sup>3</sup>	9221CE <sup>4</sup>	<b>D</b> 00 <b>F</b> 0 0 <b>F</b>	
	membrane filter $(MF)^2$ , single step.	p.124 <sup>3</sup>	9222D <sup>4</sup>	B-0050-85 <sup>5</sup>	
2. Coliform (fecal) in presence of chlorine number per 100 ml	MPN, 5 tube, 3 dilution; or MF, single step <sup>6</sup>	p.132 <sup>3</sup> p.124 <sup>3</sup>	$9221CE^4$ $9222D^4$		
3. Coliform (total) number per 100 ml	MPN, 5 tube, 3 dilution; or, MF <sup>2</sup> single step or two step	p.114 <sup>3</sup> p.108 <sup>3</sup>	$9221B^4\\9222B^4$	B-0025-85 <sup>5</sup>	
4. Coliform (total) in presence of chlorine, number per 100 ml	MPN, 5 tube, 3 dilution; or, MF <sup>2</sup> with enrichment.	p.114 <sup>3</sup> p.111 <sup>3</sup>	$9221B^4$ $9222(B+B.5c)^4$		
5. <i>E. coli</i> , number per $100 \mathrm{mL}^{28}$ .	MPN <sup>7,9,15</sup> , multiple tube; Multiple tube/multiple well;		$\begin{array}{c} 9221B.1/9221F^{4,12,14}\\ 9223B^{4,13} \end{array}$		991.15 <sup>11</sup> Colilert® <sup>13,17</sup>
	$\mathrm{MF}^{2,6,7.8,9}$ two step, or	1103.1 <sup>20</sup>	9222B/9222G <sup>4,19</sup> 9213D <sup>4</sup>		Colilert-18® <sup>13,16,17</sup> D5392-93 <sup>10</sup>
	Single step.	$\frac{1603^{21}}{1604^{22}}$			MColiBue 24 <sup>18</sup>
6. Fecal streptococci, number per 100 ml	MPN, 5 tube, 3 dilution; MF <sup>2</sup> , or	p.139 <sup>3</sup> p.136 <sup>3</sup>	9230B <sup>4</sup>	B-0055-85 <sup>5</sup>	
100 m	Plate count	p.143 <sup>4</sup>		2 0000 00	
7. Enterococci, number per 100 mL.	MPN <sup>7.9</sup> multiple tube. Multiple tube/multiple well.		9230B <sup>4</sup>		D6503-99 <sup>10</sup>
	MF <sup>2.6,789</sup> two step. Single step, or	$\frac{1106.1^{24}}{1600^{25}}$	9230C <sup>4</sup>		Enterolert® <sup>13,23</sup> D5259-92 <sup>10</sup>
	Plate count	p. 143 <sup>3</sup>			
Protozoa:					
8. Cryptosporidium <sup>28</sup>	Filtration/IMS/FA	$1622^{26}$ $1623^{27}$			
9. Giardia <sup>28</sup>	Filtration/IMS/FA	1623 <sup>27</sup>			
Aquatic Toxicity:					
10. Toxicity, acute, fresh water organisms, percent effluent	Ceriodaphnia, 48-h static-renewal mortality.				Note 29
	Fathead minnow, 96-h static- renewal mortality, or 96-h flow- through mortality.				Note 29
11. Toxicity, chronic, fresh water organisms, percent effluent.	Fathead minnow larval survival and growth.				Note 29
	Ceriodaphnia survival and reproduction.				Note 29

# TABLE A LIST OF APPROVED BIOLOGICAL TEST PROCEDURES

SECTION 3. NR 219.04 TABLE A Notes are repealed and recreated to read:

#### TABLE A NO TES:

- <sup>1</sup> The method must be specified when results are reported.
- $^{2}$  A 0.45  $\mu$ m membrane filter (MF) or other pore size certified by the manufacturer to fully retain organisms to be cultivated and to be free of extractables which could interfere with their growth.
- <sup>3</sup> USEPA. 1978. Microbiological Methods for Monitoring the Environment, Water, and Wastes. Environmental Monitoring and Support Laboratory, U.S. Environmental Protection Agency, Cincinnati, Ohio. EPA/600/8–78/017.
- <sup>4</sup> APHA. 1998, 1995, 1992. Standard Methods for the Examination of Water and Wastewater. American Public Health Association. 20th, 19th, and 18th Editions. Amer. Publ. Hlth. Assoc., Washington, D.C.
- <sup>5</sup> USGS. 1989. U.S. Geological Survey Techniques of Water-Resource Investigations, Book 5, Laboratory Analysis, Chapter A4, Methods for Collection and Analysis of Aquatic Biological and Microbiological Samples, U.S. Geological Survey, U.S. Department of Interior, Reston, Virginia.
- <sup>6</sup> Because the MF technique usually yields low and variable recovery from chlorinated wast ewaters, the Most Probable Number method will be required to resolve any controversies.
- <sup>7</sup> Tests must be conducted to provide organism enumeration (density). Select the appropriate configuration of tubes/filtrations and dilutions/volumes to account for the quality, character, consistency, and anticipated organism density of the water sample.
- <sup>8</sup> When the MF method has not been used previously to test ambient waters with high turbidity, large number of noncoliform bacteria, or samples that may contain organisms stressed by chlorine, a parallel test should be conducted with a multiple-tube technique to demonstrate applicability and comparability of results.
- <sup>9</sup> To assess the comparability of results obtained with individual methods, it is suggested that side-by-side tests be conducted across seasons of the year with the water samples routinely tested in accordance with the most current Standard Methods for the Examination of Water and Wastewater or EPA alternate test procedure (ATP) guidelines.
- <sup>10</sup> AST M. 2000, 1999, 1996. Annual Book of AST M Standards—Water and Environmental Technology. Section 11.02. American Society for Testing and Materials. 100 Barr Harbor Drive, West Conshohocken, PA 19428.
- <sup>11</sup> AOAC. 1995. Official Methods of Analysis of AOAC International, 16th Edition, Volume I, Chapter 17. Association of Official Analytical Chemists International. 481 North Frederick Avenue, Suite 500, Gaithersburg, Maryland 20877–2417.
- <sup>12</sup> The multiple-tube fermentation test is used in 9221B.1. Lactose broth may be used in lieu of lauryl tryptose broth (LTB), if at least 25 parallel tests are conducted bet ween this broth and LTB using the water samples normally tested, and this comparison demonstrates that the false-positive rate and false-negative rate for total coliform using lactose broth is less than 10 percent. No requirement exists to run the completed phase on 10 percent of all total coliform-positive tubes on a seasonal basis.
- <sup>13</sup> These tests are collectively known as defined enzyme substrate tests, where, for example, a substrate is used to detect the enzyme bglucuronidase produced by *E. coli*.
- <sup>14</sup> After prior enrichment in a presumptive medium for total coliform using 9221B.1, all presumptive tubes or bottles showing any amount of gas, growth or acidity within 48 h ± 3 h of incubation shall be submitted to 9221F. Commercially available EC–MUG media or EC media supplemented in the laboratory with 50 g/mL of MUG may be used.
- <sup>15</sup> Samples shall be enumerated by the multiple-tube or multiple-well procedure. Using multiple-tube procedures, employ an appropriate tube and dilution configuration of the sample as needed and report the Most Probable Number (MPN). Samples tested with Colilert may be enumerated with the multiple-well procedures, Quanti-Tray or Quanti-Tray 2000, and the MPN calculated from the table provided by the manufacturer.
- <sup>16</sup> Colilert-18 is an optimized formulation of the Colilert for the determination of total coliforms and *E. coli* that provides results within 18 h of incubation at 35°C rather than the 24 h required for the Colilert test and is recommended for marine water samples.
- <sup>17</sup> Descriptions of the Colilert, Colilert-18, Quanti-Tray, and Quanti-Tray/2000 may be obtained from IDEXX Laboratories, Inc., One IDEXX Drive, Westbrook, Maine 04092.
- <sup>18</sup> A description of the mColiBlue24" test, Total Coliforns and E. coli, is available from: Hach Company, 100 Dayton Ave., Ames, IA 50010.
- <sup>19</sup> Subject total coliform positive samples determined by 9222B or other membrane filter procedure to 9222G using NA–MUG media.
- <sup>20</sup> USEPA. 2002. Method 1103.1: Escherichia coli (E. coli) In Water By Membrane Filtration Using membrane -Thermotolerant Escherichia coli Agar (mTEC). U.S. Environmental Protection Agency, Office of Water, Washington D.C. EPA-821-R-02-020.
- <sup>21</sup> USEPA. 2002. Method 1603: Escherichia coli (E. coli) In Water By Membrane Filtration Using Modified membrane -Thermotolerant Escherichia coli Agar (modified mTEC). U.S. En vironmental Protection Agency, Office of Water, Washington D.C. EPA-821-R-02-023.

- <sup>22</sup> Preparation and use of MI agar with a standard membrane filter procedure is set forth in the article, Brenner et al. 1993. "New Medium for the Simultaneous Detection of Total Coliform and Escherichia coli in Water." Appl. Environ. Microbiol. 59:3534-3544 and in USEPA. 2002. Method 1604: Total Coliforms and Escherichia coli (E. coli) in Water by Membrane Filtration by Using a Simultaneous Detection Technique (MIMedium). U.S. Environmental Protection Agency, Office of Water, Washington DC. EPA 821-R-02-024.
- <sup>23</sup> A description of the Enterolert [reg] test may be obtained from IDEXX Laboratories, Inc., One IDEXX Drive, Westbrook, Maine 0 4092.
- <sup>24</sup> USEPA. 2002. Method 1106.1: Enterococci In Water By Membrane Filtration Using membrane-Enterococcus-Esculin Iron Agar (mE-EIA). U.S. Environmental Protection Agency, Office of Water, Washington DC. EPA-821-R-02-021.
- <sup>25</sup> USEPA. 2002. Method 1600: Enterococci in Water by Membrane Filtration Using membrane-Enterococcus Indoxyl-[beta]-D-Glucoside Agar (mEI). U.S. Environmental Protection Agency, Office of Water, Washington, DC. EPA-821-R-02-022.
- <sup>26</sup> Method 1622 uses filtration, concentration, immunomagnetic separation of oocysts from captured material, immunofluorescence assay to determine concentrations, and confirmation through vital dye staining and differential interference contrast microscopy for the detection of Cryptosporidium. SEPA. 2001. Method 1622: Cryptosporidium in Water by Filtration/IMS/FA. U.S. Environmental Protection Agency, Office of Water, Washington DC. EPA-821-R-01-026.
- <sup>27</sup> Method 1623 uses filtration, concentration, immunomagnetic separation of oocysts and cysts from captured material, immunofluorescence assay to determine concentrations, and confirmation through vital dye staining and differential interference contrast microscopy for the simultaneous detection of Cryptosporidium and Giardia oocysts and cysts. USEPA. 2001. Method 1623. Cryptosporidium and Giardia in Water by Filtration/IMS/FA. U.S. Environmental Protection Agency, Office of Water, Washington DC. EPA-821-R-01-025.
- <sup>28</sup> Recommended for enumeration of target organism in ambient water only.
- <sup>29</sup> Compliance monitoring must be performed in accordance with the specifications in the "State of Wisconsin Aquatic Life Toxicity Testing Methods Manual, 1st Edition," Wisconsin Department of Natural Resources, 1996. This publication is available for inspection at the offices of the Department of Natural Resources, the Secretary of State and the Revisor of Statutes. Copies are Available from: the Department of Natural Resources, Bureau of Integrated Science Services, P.O. Box 7921, Madison, W153707.

# SECTION 4. NR 219.04 TABLE B is repealed and recreated to read:

	LIST OF APPROVED I			Standard Methods <sup>3p</sup>			
ira	meter, Units & Methods	EPA <sup>1</sup>	SW-846 <sup>3c,3i</sup>		ASTM <sup>3w</sup>	USGS <sup>2</sup>	Other
	Acidity, as CaCO <sub>3</sub> , mg/L, Electrometric end point or phenolphthalein end point	305.1		2310 B(4a) [18 <sup>th</sup> , 19 <sup>th</sup> , 20 <sup>th</sup> ]	D1067-92	I-1020-85	
	Alkalinity, as CaCO <sub>3</sub> , mg/L; Electrometric or colorimetric:						
	Titration to pH 4.5, manual or automated	310.1 310.2		2320 B [18th, 19th, 20th]	D1067-92	I-1030-85 I-2030-85	973.43 <sup>3</sup>
	Aluminum, mg/L: Digestion <sup>4</sup> followed by:	202.1	7020			1 2051 05	
	AA direct aspiration <sup>4m,36</sup> , AA furnace,	202.1 202.2 or 200.9 <sup>5</sup>	7020	3111 D [18th, 19th] 3113 B [18th, 19th]		I-305I-85	
	Inductively coupled plasma (ICP) <sup>4m,36</sup> , Inductively coupled plasma-mass spectrometry (ICP-MS), Direct current plasma (DCP) <sup>4m,36</sup> , or	$200.7^{5}$ $200.8^{5}$	6010C 6020A	3120 B [18th, 19th, 20th]	D4190-94	I-4471-97 <sup>50</sup>	Note 34
	Colorimetric (Eriochrome cyanine R)			3500-Al B [20th] or 3500- Al D [18th, 19th]			
	Ammonia (as N), mg/L: Manual distillation <sup>6</sup> (at pH 9.5):	350.2		4500-NH <sub>3</sub> B [18th, 19th, 20th]			
	Followed by			2011]			973.49
	Nesslerization, Titration,	350.2 350.2		4500-NH <sub>3</sub> C [18 <sup>th</sup> ] 4500-NH <sub>3</sub> C [19th, 20th] and 4500-NH <sub>3</sub> E [18 <sup>th</sup> ]	D1426-89(A)	I-3520-85	973.49
	Electrode,	350.3		4500-NH <sub>3</sub> D or E [19th, 20th] and 4500-NH <sub>3</sub> F or G	D1426-89(B)		
	Automated phenate, or	350.1 <sup>1m</sup>		[18th] 4500-NH <sub>3</sub> G [19th, 20th] and 4500-NH <sub>3</sub> H [18th]		I-4523-85	
	Automated electrode						Note 7
	Antimony, ug/L:						
	Digestion <sup>4</sup> followed by: AA direct aspiration <sup>4m,36</sup> ,	204.1	7040	3111 B [18th, 19th]			
	AA furnace,	200.9 <sup>5</sup>	7040	3113 B [18th, 19th]			
	AA (gaseous borohydride),		7062				
	Inductively coupled plasma <sup>4m,36</sup> , or Inductively coupled plasma-mass spectrometry	200.7 <sup>5</sup> 200.8 <sup>5</sup>	6010C 6020A	3120 B [18th, 19th, 20th]			
	Arsenic, ug/L:	206.5					
	Digestion <sup>4</sup> followed by AA (gaseous hydride), AA (gaseous borohydride),	206.5	7061A 7062	$3114 \ B^{4g} \left[ 18^{th}, 19^{th}, 20^{th} \right]$	D2972-97(B)	I-3062-85	
	AA furnace,	206.2 or 200.9 <sup>5</sup>	7060A	3113 B [18th, 19th]	D2972-97(C)	I-4063-98 <sup>49</sup>	
	Inductively coupled plasma <sup>4m,36</sup> , Inductively coupled plasma-mass spectrometry, or	$200.7^{5}$ $200.8^{5}$	6010C 6020A	3120 B [18th, 19th, 20th]	D0070 07(1)	1 20/0 07	
	Colorimetric (SDDC)			3500-As B [20th] and 3500-As C [18th, 19th]	D2972-97(A)	I-3060-85	
	Barium, mg/L:						
	Digestion <sup>4</sup> followed by: AA direct aspiration <sup>4m,36</sup> ,	208.1	7080A	3111 D [18th, 19th]		I-3084-85	
	AA furnace,	208.2	7080A 7081	3113 B [18th, 19th]	D4382-95	1-500-1-05	
	Inductively coupled plasma <sup>4m,36</sup> , Inductively coupled plasma-mass spectrometry, or Direct current plasma <sup>4m,36</sup>	200.7 <sup>5</sup> 200.8 <sup>5</sup>	6010C 6020A	3120 B [18th, 19th, 20th]			Nets 2
	Beryllium, mg/L:						Note 34
	Digestion <sup>4</sup> followed by:	210.4	7000		Darra	1 2005 05	
	AA direct aspiration,	210.1	7090	3111 D [18th, 19th]	D3654- 93(88)(A)	I-3095-85	
	AA furnace,	210.2, or	7091	3113 B [18th, 19th]	D3645-		

TABLE B LIST OF APPROVED INORGANIC TEST PROCEDURES FOR WASTEWATER

Para	meter, Units & Methods	EPA <sup>1</sup>	SW-846 <sup>3c,3i</sup>	Standard Methods <sup>3p</sup>	ASTM <sup>3w</sup>	USGS <sup>2</sup>	Other
	Inductively coupled plasma, Inductively coupled plasma-mass spectrometry	$200.7^5$ $200.8^5$	6010C 6020A	3120 B [18th, 19th, 20th]		I-4471-97 <sup>50</sup>	N. (. 24
	Direct current plasma, or Colorimetric (aluminon)			3500-Ве D [18th, 19th, 20th]	D4190-94		Note 34
9.	Biochemical oxygen demand (BOD <sub>5</sub> ), mg/L: Dissolved Oxygen Depletion			5210 B [18th, 19th, 20th]		I-1578-78 <sup>8</sup>	973.443 <sup>3</sup>
10.	Boron <sup>37</sup> , mg/L: Colorimetric (curcumin), Inductively coupled plasma, or	212.3 200.7 <sup>5</sup>	6010C	4500-B B [18th, 19th, 20th] 3120 B [18th, 19th, 20th]		I-3112-85 I-4471-97 <sup>50</sup>	
	Direct current plasma				D4190-94		Note 34
11.	Bromide, mg/L: Titrimetric	320.1			D1246-95 (C)	I-1125-85	p.S44 <sup>10</sup>
	Ion Chromatography	300.0 <sup>1m</sup>	9056				
12.	Cadmium-Total <sup>6</sup> , mg/L: Digestion <sup>4</sup> followed by:						
	AA direct aspiration4m,36,	213.1	7130	3111 B or C [18th, 19th]	D3557-95 (A or B)	I-3135-85 or I-3136-85	974.27 <sup>3</sup>
	AA furnace,	213.2, or 200.9 <sup>5</sup>	7131A	3113 B [18th, 19th]	D3557-95(D) I-4138-89		
	Inductively coupled plasma <sup>4m,36</sup>	200.75	6010C	3120 B [18th, 19th, 20th]		I-1472-85 or I-4471-97 <sup>50</sup>	
	Inductively coupled plasma-mass spectrometry Direct current plasma <sup>4m,36</sup> , Voltametry <sup>11</sup> , or Colorimetric (Dithizone)	200.85	6020A	3500-Cd D [18th, 19th]	D4190-94 D3557-95(C)		Note 34
13.	Calcium, mg/L: Digestion <sup>4</sup> followed by:						
	Atomic absorption, Inductively coupled plasma, Direct current plasma, or	215.1 200.7 <sup>5</sup>	7140 6010C	3111 B [18th, 19th] 3120 B [18th, 19th, 20th]	D511-92(B)	I-3152-85	Note 34
	EDTA titration	215.2		3500-Ca B [20th] and 3500-Ca D [18th, 19th]	D511-93(A)		100001
14.	Carbonaceous Biochemical oxygen demand (CBOD <sub>5</sub> ), mg/L: with nitrification inhibitor <sup>12</sup>			5210 B [18th, 19th, 20th]			
15.	Chemical oxygen demand (COD), mg/L: Closed reflux Titrimetric			5220 C [18th, 19th, 20th]			
		410.1 410.2 410.3		5220 B [18th, 19th, 20th]	D1252-95(A)	I-3560 or I- 3562-85	973.46 <sup>3</sup>
	Automated and manual Spectrophotometric	410.4 <sup>1m</sup>		5220 D [18th, 19th, 20th]	D1252-95(B)	I-3561-85	Notes 13 & 14
16.	Chloride, mg/L: Titrimetric (silver nitrate) or		9253	4500-Cl <sup>-</sup> B [18th, 19th,	D512-89(B)	I-1183-85	
	(Mercuric nitrate),	325.3	9252A	20th] 4500-Cl <sup>-</sup> C [18th, 19th,	D512-89(A)	I-1184-85	973.51 <sup>3</sup>
	Colorimetric (ferricyanide), manual or automated, or	325.1 or 325.2	9250	20th] 4500-Cl <sup>-</sup> E [18th, 19th, 20th]		I-1187-85 I-2187-85	
	Ion chromatography	300.0 <sup>1m</sup>	9056				
17.	Chlorine - Total residual, mg/L: amperometric,	330.1		4500-Cl D [18th, 19th,	D1253-86(92)		
	Starch End point direct	330.3		20th] 4500-Cl B [18th, 19th, 20th]			
	Back Titration either end point <sup>15</sup> , or	330.2		4500-Cl C [18th, 19th,			

TABLE B LIST OF APPROVED INORGANIC TEST PROCEDURES FOR WASTEWATER

Para	meter, Units & Methods	EPA <sup>1</sup>	SW-846 <sup>3c,3i</sup>	Standard Methods <sup>3p</sup>	ASTM <sup>3w</sup>	USGS <sup>2</sup>	Other
	DPD-FAS,	330.4		20th] 4500-Cl F [18th, 19th, 20th]			
	Spectrophotometric, DPD; or	330.5		4500-Cl G [18th, 19th, 20th]			
	Electrode			2011			Note 16
8.	Chromium VI dissolved, ug/L: 0.45 micron filtration with:						
	Extraction and atomic absorption,	218.4	7197	3111 C [18th, 19th]		I-1232-85	
	Coprecipitation and atomic absorption, Differential pulse polarography,		7195 7198				
	Colorimetric (Diphenylcarbazide), or		7196A	3500-Cr B [20th] and 3500-Cr D [18th, 19th]	D1687-92(A)	I-1230-85	
	Ion Chromatography	218.6 <sup>5</sup>					
9.	Chromium, mg/L: Digestion <sup>4</sup> (optional extraction)						
	followed by: AA direct aspiration <sup>4m,36</sup> ,	218.1	7190	3111 B [18th, 19th]	D1687-92(B)	I-3236-85	974.27
	AA chelation extraction, AA furnace,	218.3 218.2, or	7191	3111 C [18th, 19th] 3113B [18th, 19th]	D1687-92(C)	I-3233-93 <sup>46</sup>	
	Inductively coupled plasma <sup>4m,36</sup> , Inductively coupled plasma-mass spectrometry,	$200.9^{5}$ $200.7^{5}$ $200.8^{5}$	6010C 6020A	3120B [18th, 19th, 20th]			
	Direct current plasma <sup>4m,36</sup> , or	200.8	0020A		D4190-94		Note 3
	Colorimetric (diphenylcarbazide),			3500-Cr B [20th] and 3500-Cr D [18th, 19th]			
0.	Cobalt, mg/L:						
	Digestion <sup>4</sup> followed by: AA direct aspiration,	219.1	7200	3111 B [18th, 19th]	D3558-94(A or	I-3239-85	
	AA furnace, or	219.2, or	7201	3113 B [18th, 19th]	B) D3558-94(C)	I-4243-89 <sup>51</sup>	
	Inductively coupled plasma, or	$200.9^{5}$ $200.7^{5}$	6010C	3120 B [18th, 19th, 20th]		I-4471-97 <sup>50</sup>	
	Inductively coupled plasma-mass spectrometry Direct current plasma	200.85	6020A		D4190-94		Note 3-
1.	Color, PlatinumCobalt units or						
	dominant wavelength hue, luminance, purity:						
	Colorimetric, ADMI	110.1		2120 E [18th, 19th, 20th]			Note 1
	Platinumcobalt; or Spectrophotometric	110.2 110.3		2120 B [18th, 19th, 20th] 2120 C [18th, 19th, 20th]		I-1250-85	
2		110.5		2120 C [1800, 1900, 2000]			
2.	Copper, mg/L: Digestion <sup>4</sup> followed by:						
	AA direct aspiration <sup>4m,36,</sup>	220.1	7210	3111 B or C [18th, 19th]	D1688-95(A or B)	I-3270-85	974.27
	AA furnace,	220.2 or 200.9 <sup>5</sup>	7211	3113 B [18th, 19th]	D1688-95(C)	I-4274-89 <sup>51</sup>	
	Inductively coupled plasma <sup>4m,36</sup> Inductively coupled plasma-mass spectrometry	$200.7^5$ $200.8^5$	6010C 6020A	3120 B [18th, 19th, 20th]		I-4471-97 <sup>50</sup>	
	Direct current plasma <sup>4m,36</sup> , Colorimetric (Neocuproine), or			3500-Cu B [20th] and	D4190-94		Note 3
	Bicinchoninate			3500-Cu D [20th] and 3500-Cu D [18th, 19th] 3500-Cu C [20th]			Note 1
2				-			
3.	Cyanide - Total, ug/L: Manual distillation with MgCl <sub>2</sub>			4500-CN-C [18th, 19th, 20th]	D2036-98(A)		
	Followed by: titrimetric,			4500-CN-D [18th, 19th, 20th]			
	Manual or	335.2 <sup>31</sup>	9010C	4500-CN-E [18th, 19th, 20th]	D2036-98(A)	I-3300-85	
	Automated <sup>20</sup> spectrophotometric, or Semi-automated colorimetry	335.3 <sup>31</sup> 335.4 <sup>1m</sup>	9010C 9012B	j		I-4302-85	
	atomated coronnally	000.4	> • · • • •				

24. Cyanide, Available, ug/L:

Para	meter, Units & Methods	EPA <sup>1</sup>	SW-846 <sup>3c,3i</sup>	Standard Methods <sup>3p</sup>	ASTM <sup>3w</sup>	USG S <sup>2</sup>	Other
	Cyanide amenable to chlorination (CATC), Manual distillation with MgCl <sub>2</sub> followed by titrimetric, manual or automated spectrophotometric Flow injection and ligand exchange, followed by amperometry	335.1	9010C	4500-CN-G [18th, 19th, 20 <sup>th</sup> ]	D2036-98(B)		OIA-1677 <sup>4</sup>
25.	Fluoride - Total, mg/L: Manual distillation <sup>8</sup> Followed by manual or automated electrode,	340.2		4500-F-B [18th, 19th, 20th] 4500-F-C [18th, 19th, 20th]	D1179-93(B)	I-4327-85	
	SPADNS,	340.1		4500-F-D [18th, 19th, 20th]	D1179-93(A)	1-4527-05	
	Ion chromatography, or automated complexone	300.0 <sup>1m</sup> 340.3	9056	4500-F-E [18th, 19th, 20th]			
26.	Gold, mg/L: Digestion <sup>4</sup> followed by: AA direct aspiration AA furnace, Direct current plasma, or	231.1 231.2 200.7 <sup>5</sup>	6010C	3111 B [18th, 19th] 3113 B [18th, 19th]			Note 34
27.	Inductively coupled plasma Hardness - Total as CaCO <sub>3</sub> , mg/L: Automated colorimetric,	130.1	8010C				
	EDTA titration, or the sum of Ca and Mg as their respective carbonates (by ICP or AA direct aspiration) (See Parameters 13 and 33)	130.2		2340 B or C [18th, 19th, 20th]	D1126-86(92)	I-1338-85	973.52B <sup>3</sup>
28.	Hydrogen ion (pH), pH units: Electrometric Measurements or	150.1	9040C	4500-H+B [18th, 19th, 20th]	D1293-84(90) (A or B)	I-1586-85	973.41 <sup>3</sup>
	Automated Electrode			2011	(1101 D)	I-2587-85	Note 21
29.	Iridium, ug/L: Digestion <sup>4</sup> followed by: AA direct aspiration, AA furnace, or Inductively coupled plasma	235.1 235.2 200.7 <sup>5</sup>	6010C	3111 B [18th, 19th]			
30.	Iron, mg/L: Digestion <sup>4</sup> followed by: AA direct aspiration <sup>4m.36</sup> ,	236.1	7380	3111 B or C [18th, 19th]	D1068-96	I-3381-85	974.27 <sup>3</sup>
	AA furnace,	236.2 or	7381	3113 B [18th, 19th]	(A or B) D1068-96(C)	10001 00	<i>,,</i> ,
	Inductively coupled plasma <sup>4m, 36</sup> ,	$200.9^{5}$ $200.7^{5}$	6010C	3120 B [18th, 19th, 20th]		I-4471-97 <sup>50</sup>	
	Direct current plasma <sup>4m,36</sup> , or Colorimetric (Phenanthroline)			3500-Fe B [20th] and 3500-Fe D [18th, 19th]	D4190-94 D1068-96(D)		Note 34 Note 22
31.	Kjeldahl nitrogen - Total (as N), mg/L: Digestion and distillation	351.3		4500-N <sub>org</sub> B or C [18th, 19th, 20th] and 4500-NH <sub>3</sub> B [18th, 19th, 20th]	D3590-89(A)		
	Followed by titration Nesslerization or Electrode,	351.3 351.3 351.3		4500-NH <sub>3</sub> C [18th] 4500-NH <sub>3</sub> C [19th, 20th] and 4500-NH <sub>3</sub> E [18th]	D3590-89(A) D3590-89(A)	1 4551 708	937.46 <sup>3</sup>
	Automated phenate, Semi-automated block digester,	351.1 351.2 <sup>1m</sup>		4500-NH <sub>3</sub> G [18th, 19th, 20th]	D3590-89(B)	I-4551-78 <sup>8</sup> I-4515-91 <sup>45</sup>	
	or potentiometric Block digester, followed by Auto distillation and Titration,	351.2			D3590-89(A)	010 /1	Note 39
	or Nesslerization, or						Note 40

TABLE B LIST OF APPROVED INORGANIC TEST PROCEDURES FOR WASTEWATER

Parai	neter, Units & Methods	EPA <sup>1</sup>	SW-846 <sup>3c,3i</sup>	Standard Methods <sup>3p</sup>	ASTM <sup>3w</sup>	USG S <sup>2</sup>	Other
		-					
	Flow injection gas diffusion						Note 41
2.	Lead, mg/L:						
	Digestion <sup>4</sup> followed by: $A = A^{1/2} + A^{$	220.1	7420	2111 D C [10:1 10:1]	D2550.00	1 2200 00	074 073
	AA direct aspiration <sup>4m,36</sup> ,	239.1	7420	3111 B or C [18th, 19th]	D3559-90 (A or B)	I-3399-90	974.27 <sup>3</sup>
	AA furnace,	239.2 or	7421	3113 B [18th, 19th]	D3559-90(C)	I-4403-89 <sup>51</sup>	
		200.9 <sup>5</sup>				50	
	Inductively coupled plasma <sup>4m,36</sup> ,	200.75	6010C	3120 B [18th, 19th, 20th]		I-4471-97 <sup>50</sup>	
	Inductively coupled plasma-mass spectrometry Direct current plasma <sup>4m,36</sup> ,	200.85	6020A		D4190-94		Note 34
	Voltametry <sup>11</sup> or				D3559-90(C)		Note 34
	Colorimetric (Dithizone)			3500-Pb B [20th] and			
				3500-Pb D [18th, 19th]			
3.	Magnesium, mg/L:						
	Digestion <sup>4</sup> followed by:						
	Atomic absorption,	242.1	7450	3111 B [18th, 19th]	D511-93(B)	I-3447-85	974.27 <sup>3</sup>
	Inductively coupled plasma, Direct current plasma, or	200.7 <sup>5</sup>	6010C	3120 B [18th, 19th, 20th]		I-4471-97 <sup>50</sup>	Note 34
	Gravimetric			3500-Mg D [18th, 19th]			11010 34
				22.50 mg 2 [10m, 19m]			
ŀ.	Manganese, mg/L: Digestion <sup>4</sup> followed by:						
	AA direct aspiration <sup>4m,36</sup> ,	243.1	7460	3111 B [18th, 19th]	D858-90	I-3454-85	974.27
	<b>1</b> ,			. , 1	(A or B)		
	AA furnace,	243.2 or	7461	3113 B [18th, 19th]	D858-90(C)		
	Inductively coupled plasma <sup>4m,36</sup> ,	$200.9^{5}$ $200.7^{5}$	6010C	3120 B [18th, 19th, 20th]		I-4471-97 <sup>50</sup>	
	Inductively coupled plasma mass spectrometry,	200.7 200.8 <sup>5</sup>	6020A	5120 B [18th, 19th, 20th]		1-44/1-9/	
	Direct current plasma <sup>4m,36</sup> ,	200.0	002011		D4190-94		Note 34
	Colorimetric (Persulfate), or			3500-Mn B [20th] and			920.203
	Periodate			3500-Mn D [18th, 19th]			Note 23
i.	Mercury - Total <sup>6</sup> , ug/L:						
	Cold vapor AA, manual or	$245.1^{5}$	7470A	3112 B [18th, 19th, 20th]	D3223-91	I-3462-85	977.22
	automated, or	245.2					
	Oxidation, purge and trap, cold vapor atomic fluorescence spectrometry (ng/l) <sup>43m</sup>	1631E <sup>43</sup>					
	Oxidation, cold vapor atomic fluorescence spectrometry	245.7 <sup>5</sup>					
	(ng/l) <sup>43m</sup>						
5m.	Mercury - Hg(II) and organomercurials, ug/L:						
	HPLC with electrochemical detection	245.3 <sup>5</sup>					
	Molybdenum, mg/L:						
	Digestion <sup>4</sup> followed by:						
	AA direct aspiration,	246.1	7480	3111 D [18th, 19th]		I-3490-85	
	AA furnace,	246.2	7481	3113 B [18th, 19th]		I-3492-96 <sup>47</sup>	
	Inductively coupled plasma, Inductively coupled plasma-mass spectrometry, or	$200.7^{5}$ $200.8^{5}$	6010C 6020A	3120 B [18th, 19th, 20th]		I-4471-97 <sup>50</sup>	
	Direct current plasma	200.8	0020A				Note 34
	-						
	Nickel, mg/L: Digestion <sup>4</sup> followed by:						
	Digestion followed by: AA direct aspiration $4^{m,36}$ ,	249.1	7520	3111 B or C [18th, 19th]	D1886-90	I-3499-85	
	····	- (7.1			(A or B)	00	
	AA furnace,	249.2 or		3113 B [18th, 19th]	D1886-90(C)	I-4503-89 <sup>51</sup>	
	4m.36	200.9 <sup>5</sup>	(0100	2120 D [10:1 10:1 20:1		T 4471 0750	
	Inductively coupled plasma <sup>4m,36</sup> , Inductively coupled plasma-mass spectrometry,	$200.7^{5}$ $200.8^{5}$	6010C 6020A	3120 B [18th, 19th, 20th]		I-4471-97 <sup>50</sup>	
	Direct current plasma <sup>4m,36</sup> , or	200.0	0020A		D4190-94		Note 34
	Colorimetric (Heptoxime)			3500-Ni D [17th]			
	Nitanta (ag N), mg/L						
•	Nitrate (as N), mg/L: Brucine sulfate, or	352.1					973.50
	Ditterite sufficie, of	552.1					$419D^{17}$
	Nitrate-nitrite N minus Nitrite N						

TABLE B						
LIST OF APPROVED INORGANIC TEST PROCEDURES FOR WASTEWAT	ER					

	meter, Units & Methods	EPA <sup>1</sup>	SW-846 <sup>3c,3i</sup>	Standard Methods <sup>3p</sup>	ASTM <sup>3w</sup>	USG S <sup>2</sup>	Other
	, 10	-					
	(see parameters 39 and 40) Ion chromatography <sup>25m</sup>	300.0 <sup>1m</sup>	9056				
9.	Nitrate-nitrite (as N), mg/L: Cadmium reduction, manual	353.3		4500-NO <sub>3</sub> <sup>-</sup> E [18th, 19th, 20th]	D3867-99(B)		
	or automated, or	353.2 <sup>1m</sup>		4500-NO <sub>3</sub> <sup>-</sup> F [18th, 19th, 20th]	D3867-99(A)	I-4545-85	
	automated hydrazine	353.1		4500-NO <sub>3</sub> <sup>-</sup> H [18th, 19th, 20th]			
	Ion chromatography <sup>25m</sup>	300.0 <sup>1m</sup>	9056	2001			
0.	Nitrite (as N), mg/L: Spectrophotometric, manual or	354.1		4500-NO <sub>2</sub> <sup>-</sup> B [18th, 19th,			Note 25
	automated (Diazotization),			20th]		I-4540-85	
	Automated (bypass cadmium reduction), or	353.2 Rev. 2.0				1-4540-65	
	Ion chromatography <sup>25m</sup>	300.0 <sup>1m</sup>	9056				
1.	Oil and grease-Total recoverable, mg/L:						
	Gravimetric (freon extraction) Oil and grease and non-polar material, mg/L: Hexane extractable material (HEM): n-hexane extraction and	413.1 1664A <sup>42</sup>	9070	5520 B [18th, 19th, 20th] <sup>38</sup> 5520 B [18th, 19th, 20th] <sup>38</sup>			
	gravimetry. Silica gel treated HEM (SGT-HEM): Silica gel treatment and gravimetry.	1664A <sup>42</sup>					
2.	Organic carbon - Total (TOC), mg/L:						
	Combustion or oxidation,	415.1	9060A	5310 B, C, or D [18th,	D2579-93	p.14 <sup>24</sup>	973.47 <sup>3</sup>
	Persulfate oxidation	415.2 <sup>1m</sup>		19th, 20th] 5310 C [18th, 19th, 20th]	(A or B)		
2m.	Organic Halides, Adsorbable (AOX), ug/L Adsorption and coulometric titration	$1650^{42m}$					
3.	Organic nitrogen (as N), mg/L: Total Kjeldahl N (Parameter 31) minus ammonia N (Parameter 4)						
4.	Orthophosphate (as P), mg/L:						
	Ascorbic acid method,	365.1		4500-P F [18th, 19th, 20th]		I-4601-85	973.56 <sup>3</sup>
	automated						
	Or manual single reagent or	365.2		4500-P E [18th, 19th, 20th]	D515-88(A)		973.55 <sup>3</sup>
	Manual two reagent, or	365.3	0056				
	Ion chromatography	300.0 <sup>1m</sup>	9056				
5.	Osmium, ug/L:	300.0***	9056				
5.	Osmium, ug/L: Digestion <sup>4</sup> followed by:						
5.	Osmium, ug/L: Digestion <sup>4</sup> followed by: AA direct aspiration,	252.1	9056 7550	3111 D [18th, 19th]			
5.	Osmium, ug/L: Digestion <sup>4</sup> followed by: AA direct aspiration, AA furnace, or	252.1 252.2	7550	3111 D [18th, 19th]			
5.	Osmium, ug/L: Digestion <sup>4</sup> followed by: AA direct aspiration,	252.1		3111 D [18th, 19th]			
	Osmium, ug/L: Digestion <sup>4</sup> followed by: AA direct aspiration, AA furnace, or	252.1 252.2	7550	3111 D [18th, 19th]			
	Osmium, ug/L: Digestion <sup>4</sup> followed by: AA direct aspiration, AA furnace, or Inductively coupled plama	252.1 252.2	7550	4500-O C [18th, 19th,	D888-92(A)	I-1575-78 <sup>8</sup>	973.45 B
	Osmium, ug/L: Digestion <sup>4</sup> followed by: AA direct aspiration, AA furnace, or Inductively coupled plama Oxygen, dissolved, mg/L:	252.1 252.2 200.7 <sup>5</sup>	7550		D888-92(A) D888-92(B)	I-1575-78 <sup>8</sup> I-1576-78 <sup>8</sup>	973.45 B
5.	Osmium, ug/L: Digestion <sup>4</sup> followed by: AA direct aspiration, AA furnace, or Inductively coupled plama Oxygen, dissolved, mg/L: Winkler (Azide modification) Or electrode	252.1 252.2 200.7 <sup>5</sup> 360.2	7550	4500-O C [18th, 19th, 20th] 4500-O G [18th, 19th,			973.45 B
6.	Osmium, ug/L: Digestion <sup>4</sup> followed by: AA direct aspiration, AA furnace, or Inductively coupled plama Oxygen, dissolved, mg/L: Winkler (Azide modification) Or electrode Palladium, mg/L:	252.1 252.2 200.7 <sup>5</sup> 360.2	7550	4500-O C [18th, 19th, 20th] 4500-O G [18th, 19th,			973.45 B
6.	Osmium, ug/L: Digestion <sup>4</sup> followed by: AA direct aspiration, AA furnace, or Inductively coupled plama Oxygen, dissolved, mg/L: Winkler (Azide modification) Or electrode Palladium, mg/L: Digestion <sup>4</sup> followed by:	252.1 252.2 200.7 <sup>5</sup> 360.2 360.1	7550	4500-O C [18th, 19th, 20th] 4500-O G [18th, 19th, 20th]			
6.	Osmium, ug/L: Digestion <sup>4</sup> followed by: AA direct aspiration, AA furnace, or Inductively coupled plama Oxygen, dissolved, mg/L: Winkler (Azide modification) Or electrode Palladium, mg/L: Digestion <sup>4</sup> followed by: AA direct aspiration,	252.1 252.2 200.7 <sup>5</sup> 360.2 360.1	7550	4500-O C [18th, 19th, 20th] 4500-O G [18th, 19th,			p. S27 <sup>10</sup>
6.	Osmium, ug/L: Digestion <sup>4</sup> followed by: AA direct aspiration, AA furnace, or Inductively coupled plama Oxygen, dissolved, mg/L: Winkler (Azide modification) Or electrode Palladium, mg/L: Digestion <sup>4</sup> followed by: AA direct aspiration, AA furnace,	252.1 252.2 200.7 <sup>5</sup> 360.2 360.1	7550	4500-O C [18th, 19th, 20th] 4500-O G [18th, 19th, 20th]			p. S27 <sup>10</sup> p. S28 <sup>10</sup>
6.	Osmium, ug/L: Digestion <sup>4</sup> followed by: AA direct aspiration, AA furnace, or Inductively coupled plama Oxygen, dissolved, mg/L: Winkler (Azide modification) Or electrode Palladium, mg/L: Digestion <sup>4</sup> followed by: AA direct aspiration,	252.1 252.2 200.7 <sup>5</sup> 360.2 360.1	7550	4500-O C [18th, 19th, 20th] 4500-O G [18th, 19th, 20th]			p. S27 <sup>10</sup>
6. 7.	Osmium, ug/L: Digestion <sup>4</sup> followed by: AA direct aspiration, AA furnace, or Inductively coupled plama Oxygen, dissolved, mg/L: Winkler (Azide modification) Or electrode Palladium, mg/L: Digestion <sup>4</sup> followed by: AA direct aspiration, AA furnace, Direct current plasma, or Inductively coupled plasma	252.1 252.2 200.7 <sup>5</sup> 360.2 360.1 253.1 253.2	7550 6010C	4500-O C [18th, 19th, 20th] 4500-O G [18th, 19th, 20th]			p. S27 <sup>10</sup> p. S28 <sup>10</sup>
	Osmium, ug/L: Digestion <sup>4</sup> followed by: AA direct aspiration, AA furnace, or Inductively coupled plama Oxygen, dissolved, mg/L: Winkler (Azide modification) Or electrode Palladium, mg/L: Digestion <sup>4</sup> followed by: AA direct aspiration, AA furnace, Direct current plasma, or	252.1 252.2 200.7 <sup>5</sup> 360.2 360.1 253.1 253.2	7550 6010C	4500-O C [18th, 19th, 20th] 4500-O G [18th, 19th, 20th]			p. S27 <sup>10</sup> p. S28 <sup>10</sup>

TABLE B LIST OF APPROVED INORGANIC TEST PROCEDURES FOR WASTEWATER

Para	meter, Units & Methods	EPA <sup>1</sup>	SW-846 <sup>3c,3i</sup>	EDURES FOR WASTEWA' Standard Methods <sup>3p</sup>	ASTM <sup>3w</sup>	USG S <sup>2</sup>	Other
	or automated <sup>20</sup> colorimetric (4AAP), or	420.2	9066				Note 27
	Semi-automated colorimetric	420.2 420.4 <sup>1m</sup>	9000				Note 27
19.	Phosphorus (elemental), mg/L: Gas-Liquid chromatography						Note 28
50.	Phosphorus - Total, mg/L: Persulfate digestion	365.2		4500-P B,5 [18th, 19th, 20th]			973.53 <sup>3</sup>
	Followed by manual or	365.2 or 365.3		4500-P E [18th, 19th, 20th]	D515-88 (A)		
	Automated ascorbic acid	365.1 <sup>1m</sup>		4500-P F [18th, 19th, 20th]		I-4600-85	973.56 <sup>3</sup>
	Reduction, or semi-automated block digestor	365.4			D515-88(B)	I-4610-91 <sup>48</sup>	
51.	Platinum, mg/L:						
	Digestion <sup>4</sup> followed by:	255 1		2111 D [1945 1045]			
	AA direct aspiration, AA furnace,	255.1 255.2		3111 B [18th, 19th]			
	Direct current plasma, or	233.2					Note 34
	Inductively coupled plasma	200.7 <sup>5</sup>	6010C				
52.	Potassium, mg/L: Digestion <sup>4</sup> followed by:						
	Atomic absorption,	258.1	7610	3111 B [18th, 19th]		I-3620-85	973.53 <sup>3</sup>
	Inductively coupled plasma, Flame photometric, or	200.7 <sup>5</sup>	6010C	3120 B [18th, 19th, 20th] 3500-K B [20th] and 3500-			
	Colorimetric (cobalt nitrate)			K D [18th, 19th]			317B <sup>17</sup>
3.	Residue - total, (total solids), mg/L:						
5.	Gravimetric 103-105°C	160.3		2540 B [18th, 19th, 20th]		I-3750-85	
54.	Residue - filterable, (TDS), mg/L:						
	Gravimetric, 180°C	160.1		2540 C [18th, 19th, 20th]		I-1750-85	
55.	Residue - nonfilterable, (TSS),						
	mg/L: Gravimetric, 103-105°C post washing of residue	160.2		2540 D [18th, 19th, 20th]		I-3765-85	
56.	Residue - settleable, mg/L:	1.60.5		2540 51104 104 2041			
	Volumetric (Imhoff cone) or gravimetric	160.5		2540 F [18th, 19th, 20th]			
57.	Residue - volatile mg/L:						
	Gravimetric, 550°C	160.4				I-3753-85	
58.	Rhodium, ug/L: Digestion <sup>4</sup> followed by:						
	AA direct aspiration,	265.1		3111 B [18th, 19th]			
	AA furnace, or	265.2		,			
	Inductively coupled plasma	200.7 <sup>5</sup>	6010C				
59.	Ruthenium, ug/L:						
	Digestion <sup>4</sup> followed by: AA direct aspiration,	267.1		3111 B [18th, 19th]			
	AA direct aspiration, AA furnace, or	267.1		JIII D [1000, 1900]			
	Inductively coupled plasma	200.7 <sup>5</sup>	6010C				
50.	Selenium, ug/L:						
	Digestion <sup>4</sup> followed by: AA furnace,	270.2 or	7740	3113 B [18th, 19th]	D3859-98(B)	I-4668-98 <sup>49</sup>	
		$200.9^{5}$	1140	5115 B [1000, 1700]	23037-70( <b>D</b> )	1 -000-20	
	Inductively coupled plasma <sup>4m,36</sup> ,	200.75	6010C	3120 B [18th, 19th, 20th]			
	Inductively coupled plasma-mass spectrometry,	200.8 <sup>5</sup>	6020A	3114 B <sup>4g</sup> [18th, 19th, 20th]	D2050 00/4	12667 05	
	or AA (gaseous hydride)		7741A	5114 B ° [18th, 19th, 20th]	D3839-98(A)	I-3667-85	
51.	Silica - Dissolved, mg/L: 0.45 micron filtration:						

TABLE B LIST OF APPROVED INORGANIC TEST PROCEDURES FOR WASTEWATER

0		EPA <sup>1</sup>	SW-846 <sup>3c,3i</sup>	EDURES FOR WASTEWAT Standard Methods <sup>3p</sup>		USC 5 <sup>2</sup>	Other
ara	meter, Units & Methods	EPA*	SW-846 <sup>32,34</sup>		ASTM <sup>3w</sup>	USG S <sup>2</sup>	Other
	Followed by manual or	370.1		4500-Si C [20th] and 4500- Si D [18th, 19th]	D859-94	I-1700-85	
	automated colorimetric			51.5 [1000,1900]		I-2700-85	
	(Molybdosilicate), or	200.75	10105			T 4 4 7 4 0 7 50	
	Inductively coupled plasma <sup>6</sup>	200.75	6010C	3120 B [18th, 19th, 20th]		I-4471-97 <sup>50</sup>	
2.	Silver <sup>29</sup> , mg/L:						
	Digestion <sup>4</sup> followed by: AA direct aspiration,		7760A	2111 Por C [19th 10th]		I-3720-85	974.27 <sup>3</sup>
	AA furnace,	$200.9^{5}$	7760A 7761	3111 B or C [18th, 19th] 3113 B [18th, 19th]		I-3720-83 I-4724-89 <sup>51</sup>	974.27
	Inductively coupled plasma,	200.9 <sup>5</sup>	6010C	3120 B [18th, 19th, 20th]		I-4471-97 <sup>50</sup>	
	Inductively coupled plasma-mass spectrometry,	200.85	6020A				
	Or direct current plasma						Note 34
3.	Sodium, mg/L:						
	Digestion <sup>4</sup> followed by:						
	Atomic absorption,	273.1	7770	3111 B [18th, 19th]		I-3735-85	973.54 <sup>3</sup>
	Inductively coupled plasma, Direct current plasma, or	200.7 <sup>5</sup>	6010C	3120 B [18th, 19th, 20th]		I-4471-97 <sup>50</sup>	Note 34
	Flame photometric			3500-Na B [20th] and			
				3500-Na D [18th, 19th]			
4.	Specific conductance, micromhos/cm						
	at 25°C: Wheatstone bridge	120.1	9050A	2510 B [18th, 19th, 20th]	D1125-95(A)	I-2781-85	973.40 <sup>3</sup>
5.	Sulfate (as SO <sub>4</sub> ), mg/L:						
	Automated colorimetric	375.1	9035				
	(barium chloroanilate),	Im					
	Semi-automated colorimetric	375.2 <sup>1m</sup>	9036				
	(methylthymol blue) Gravimetric,	375.3		4500-SO4 <sup>2-</sup>			925.54
	Turbidimetria ar	375.4	9038	C or D [18th, 19th, 20th]	D516-90		426C <sup>30</sup>
	Turbidimetric, or Ion chromatography	373.4 300.0 <sup>1m</sup>	9038 9056		D310-90		420C
-	Salfda (as S) mm/L.						
5.	Sulfide (as S), mg/L: Titrimetric (iodine) or	376.1		4500-S <sup>2</sup> -F [19th, 20th] and		I-3840-85	
				4500-S <sup>2</sup> E [18th]			
	Colorimetric (methylene blue)	376.2		4500-S <sup>2-</sup> D [18th, 19th, 20th]			
_				-			
7.	Sulfite (as SO <sub>3</sub> ), mg/L: Titrimetric (iodine-iodate)	377.1		4500-S03 <sup>2-</sup> B [18th, 19th,			
	Intimetite (loume-louate)	577.1		20th]			
3.	Surfactants, mg/L: Colorimetric						
~•	(methylene blue)	425.1		5540 C [18th, 19th, 20th]	D2330-88		
9.	Temperature, °C: Thermometric	170.1		2550 B [18th, 19th, 20th]			Note 32
	-			_ , , ]			
0.	Thallium, ug/L: Digestion <sup>4</sup> followed by:						
	AA direct aspiration,	279.1	7840	3111 B [18th, 19th]			
	AA furnace,	279.2 or	7841	3113 B [18th, 19th]			
	Inductively coupled a losses of	$200.9^{5}$ $200.7^{5}$	60100	2100 D [104 104 004 ]			
	Inductively coupled plasma, or Inductively coupled plasma-mass spectrometry	$200.7^{\circ}$ $200.8^{\circ}$	6010C 6020A	3120 B [18th, 19th, 20th]			
1.	Tin, ug/L: Digestion <sup>4</sup> followed by:						
	AA direct aspiration,	282.1	7870	3111 B [18th, 19th]		I-3850-78 <sup>8</sup>	
	AA furnace, or	282.2 or		3113 B [18th, 19th]			
	In dustingly, sounded a losse	$200.9^{5}$	60100				
	Inductively coupled plasma	200.7 <sup>5</sup>	6010C				
2.	Titanium, mg/L:						
	Digestion <sup>4</sup> followed by: AA direct aspiration ,	283.1		3111 D [18th, 19th]			
	AA furnace,	283.2		3113 B [18th, 19th]			
		205.2					

TABLE B LIST OF APPROVED INORGANIC TEST PROCEDURES FOR WASTEWATER

Para	meter, Units & Methods	EPA <sup>1</sup>	SW-846 <sup>3c,3i</sup>	Standard Methods <sup>3p</sup>	ASTM <sup>3w</sup>	USGS <sup>2</sup>	Other
	Direct current plasma, or						Note 34
	Inductively coupled plasma	200.7 <sup>5</sup>	6010C				
73.	Turbidity, NTU: Nephelometric	180.1 <sup>1m</sup>		2130 B [18th, 19th, 20th]	D1889-94(A)	I-3860-85	
74.	Vanadium, mg/L:						
	Digestion <sup>4</sup> followed by:						
	AA direct aspiration,	286.1	7910	3111 D [18th, 19th]			
	AA furnace,	286.2	7911	3113 B [18th, 19th]	D3373-93		
	Inductively coupled plasma,	$200.7^{5}$	6010C	3120 B [18th, 19th, 20th]		I-4471-97 <sup>50</sup>	
	Inductively coupled plasma-mass spectrometry	$200.8^{5}$					
	Direct current plasma, or				D4190-94		Note 34
	Colorimetric (Gallic acid)			3500-V B [20th] and 3500-			
				V D [18th, 19th]			
75.	Zinc, mg/L:						
	Digestion <sup>4</sup> followed by:						
	AA direct aspiration <sup>4m,36</sup> ,	289.1	7950	3111 B or C [18th, 19th]		I-3900-85	974.27 <sup>3</sup>
	AA furnace,	289.2 or 200.9 <sup>5</sup>	7951	3113 B [18th, 19th]			
	Inductively coupled plasma <sup>4m,36</sup> ,	200.75	6010C	3120 B [18th, 19th, 20th]			
	Inductively coupled plasma-mass spectrometry,	200.8 <sup>5</sup>	6020A				
	Direct current plasma <sup>4m,36</sup> ,				D4190-94		Note 34
	Colorimetric (Dithizone), or			3500-Zn E [18th, 19th]			
	Colorimetric (Zincon)			3500-Zn B [20] and 3500-			Note 33
				Zn F [18th, 19th]			

TABLE B LIST OF APPROVED INORGANIC TEST PROCEDURES FOR WASTEWATER

SECTION 5. NR 219.04 TABLE B Notes are repealed and recreated to read:

#### TABLE B NOTES

- <sup>1</sup> "Methods for Chemical Analysis of Water and Wastes", Environmental Protection Agency, Environmental Monitoring Systems Laboratory--Cincinnati (EMSL-CI), EPA-600/4-79-020, Revised March 1983 and 1979 where applicable. Available from: National Technical Information Service, 5285 Port Royal Road, Springfield, Virginia 22161 (703) 487-4650.
- <sup>1m</sup> "Methods for the Determination of Inorganic Substances in Environmental Samples", EPA-600/R-93-100, Environmental Protection Agency, August 1993, Office of Research and Development, Washington D.C. 20460, August 1993. Available from: NTIS, 5285 Port Royal Road, Springfield, Virginia 22161 (703) 487-4650.
- <sup>2</sup> Fishman, M.J., et al. "Methods for Analysis of Inorganic Substances in Water and Fluvial Sediments", U.S. Department of the Interior, Techniques of Water-Resource Investigations of the U.S. Geological Survey, Denver, CO, Revised 1989, unless otherwise stated. Available from: U.S. Geological Survey, 604 S. Pickett Street, Alexandria, VA 22304.
- <sup>3</sup> "Official Methods of Analysis of the Association of Official Analytical Chemists", methods manual, 15th ed. (1990). Available from: The Association of Official Analytical Chemists, 1111N. 19th Street, Suite 210, Arlington, VA 22209.
- <sup>3c</sup> "T est Methods for Evaluating Solid Waste, Physical/Chemical Methods", SW-846, EPA, Office of Solid Waste and Emergency Response, 401 M Street, S.W., Washington D.C. 20460, November, 1986 (third edition), including July 1992 (update I), September 1994 (up date II), January 1995 (update IIB), December 1996 (update III), January 1998 (update IVA), November 2000 (update IVB), August 2002 (update IIIB) updates. Available from: U.S. Government Printing Office (GPO), Superintendent of Documents, Washington, DC 20402, (202) 512-1800 (Publication Number: 955-001-00000-1). Also, available on-line at http://www.epa.gov/epaoswer/hazwaste/test/sw846.htm.
- <sup>3i</sup> SW-846 series 7000 methods include SW-846 methods 7000B and 7010, the general AA method descriptions.
- <sup>3p</sup> "Standard Methods for the Examination of Water and Wastewater", Joint Editorial Board, American Public Health Association, American Water Works Association, and Water Pollution Control Federation, 20<sup>th</sup> Edition (1998), 19<sup>th</sup> Edition (1995), and 18<sup>th</sup> Edition, (1992). Available from: American Public Health Association, 1015 Fifteenth Street, N.W., Washington, D.C. 20005.
- <sup>3w</sup> "Annual Book of Standards, Section 11.01 and 11.02, Water and Environmental Technology", American Society for Testing and Materials, 1994, 1996, and 1999. Available from: the American Society for Testing and Materials, 1916 Race Street, Philadelphia, PA 19103.
- <sup>4</sup> A digestion procedure is required to solubilize suspended material and to destroy possible organic metal complexes. The required digestion procedure(s) for a particular metals analysis is listed in T able BM, Metals Digestion Procedures. Use of the graphite furnace AA technique, inductively coupled plasma, direct current plasma, as well as determination for certain elements such as arsenic, mercury, selenium, silver, and titanium require a modified digestion procedure. In all cases, the analytical method should be consulted for specific instructions and cautions.

If a digestion procedure is given in the determinative method for any of the metals in table B, and this digestion is not listed in table BM, the procedure given in the analytical method should be used; however, if the digestion included in one of the approved non-EPA references (e.g. "Standard Methods for the Examination of Water and Wastewater") is significantly different from one of the EPA procedure s listed in table BM, then the EPA procedure from table BM should be used.

Sample digestion may be omitted for AA (direct aspiration or graphite furnace), direct current plasma, and inductively coupled plasma analyses provided the sample solution to be analyzed meets the following criteria:

- (a) has a low COD ( $\overline{20}$ ),
- (b) is visibly transparent with a turbidity measurement of 1 NTU or less,
- (c) is colorless with no perceptible odor, and
- (d) is of one liquid phase and free of particulate or suspended matter following acidification.
- <sup>4g</sup> Use the digestion given in the method.
- <sup>4m</sup> "Test Methods for Evaluating Solid Waste", SW-846 method 3015A. United States EPA SW-846, 3rd Edition and updates. Footnote 3c lists the complete reference.
- <sup>5</sup> "Methods for the Determination of Metals in Environmental Samples", EPA-600/4-91-010, Environmental Protection Agency, Environmental Monitoring Systems Laboratory, Cincinnati, OH 45268, June 1991. Available from: the National Technical Information Service (NTIS), order number PB91-231498, 5258 Port Royal Road, Springfield, Virginia 22161, (703)487-4650.
- <sup>6</sup> Manual distillation is not required if comparability data on representative effluent samples are on company file to show that this preliminary distillation step is not necessary; however, manual distillation will be required to resolve any controversies.
- <sup>7</sup> Ammonia, Automated Electrode Method, Industrial Method Number 379-75 WE, dated February 19, 1976, Bran & Luebbe (Technicon) Auto Analyzer II, Bran & Luebbe Analyzing Technologies, Inc., Elmsford, NY 10523. Available from: Bran & Luebbe Analyzing Technologies, Inc., Elmsford, N.Y. 10523.
- <sup>8</sup> The approved method is that cited in "Methods for Determination of Inorganic Substances in Water and Fluvial Sediments", USGS TWRI, Book 5, Chapter A1 (1979). Available on inter-library loan.
- <sup>10</sup> "Selected Analytical Methods Approved and Cited by the United States Environmental Protection Agency", Supplement to the Fifteenth Edition of Standard Methods for the Examination of Water and Wastewater (1981). Available on inter-library loan.
- <sup>11</sup> The use of normal and differential pulse voltage ramps to increase sensitivity and resolution is acceptable.
- <sup>12</sup> Carbonaceous biochemical oxygen demand (CBOD<sub>5</sub>) must not be confused with the traditional BOD<sub>5</sub> test method which measures "total BOD". The addition of the nitrification inhibitor is not a procedural option, but must be included to report the CBOD<sub>5</sub> parameter. A discharger whose permit requires reporting the traditional BOD<sub>5</sub> may not use a nitrification inhibitor in the procedure for reporting the results. Only when a discharger's permit specifically states CBOD<sub>5</sub> is required can the permittee report data using a nitrification inhibitor.
- <sup>13</sup> OIC Chemical Oxygen Demand Method, Oceanography International Corporation, 1978, 512 West Loop, PO Box 2980, College Station, TX 77840. Available from: Oceanography International Corporation, 512 West Loop, P.O. Box 2980, College Station, TX 77840.
- <sup>14</sup> Chemical Oxygen Demand, Method 8000, Hach Handbook of Water Analysis, 1979, Hach Chemical Company, POBox 389, Loveland, CO 80537. Available from: Hach Chemical Company, P.O. Box 389, Loveland, CO 80537.
- <sup>15</sup> The back titration method will be used to resolve controversy.
- <sup>16</sup> Orion Research Instruction Manual, Residual Chlorine Electrode Model 97-70, 1977, Orion Research Incorporated, 840 Memorial Drive, Cambridge, MA 02138. Available from: Orion Research Incorporated, 840 Memorial Drive, Cambridge, MA 02138. The calibration graph for the Orion residual chlorine method must be derived using a reagent blank and three standard solutions, containing 0.2, 1.0, and 5.0 mL 0.00281 N potassium iodate/100 mL solution, respectively.
- <sup>17</sup> The approved method is that cited in Standard Methods for the Examination of Water and Wastewater, 14th Edition, 1976. Available on inter-library loan.
- <sup>18</sup> "An Investigation of Improved Procedures for Measurement of Mill Effluent and Receiving Water Color", NCASI Technical Bulletin No. 253. December, 1971. Available from: National Council of the Paper Industry for Air and Stream Improvements, Inc., 260 Madison Avenue, New York, NY 10016.
- <sup>19</sup> Copper, Biocinchoinate Method, Method 8506, Hach Handbook of Water Analysis, 1979, Hach Chemical Company, PO Box 389, Loveland, CO 80537. Available from: Hach Chemical Company, P.O. Box 389, Loveland, CO 80537.
- <sup>20</sup> After the manual distillation is completed, the autoanalyzer manifolds in EPA Methods 335.3 (cyanide) or 420.2 (phenols) are simplified by connecting the re-sample line directly to the sampler. When using the manifold setup shown in Method 335.3, the buffer 6.2 should be replaced with the buffer 7.6 found in Method 335.2.

- <sup>21</sup> Hydrogen ion (pH) Automated Electrode Method, Industrial Method Number 378-75WA, October 1976, Bran & Luebbe (Technicon) Autoanalyzer II. Bran & Luebbe Analyzing Technologies, Inc., Elmsford, NY 10523. Available from: Bran & Luebbe Analyzing Technologies, Inc. Elmsford, N.Y. 10523.
- <sup>22</sup> Iron, 1,10-Phenanthroline Method, Method 8008, 1980, Hach Chemical Company, PO Box 389, Loveland, CO 80537. Available from: Hach Chemical Company, P.O. Box 389, Loveland, CO 80537.
- <sup>23</sup> Manganese, Periodate Oxidation Method, Method 8034, Hach Handbook of Wastewater Analysis, 1979, pages 2-113 and 2-117, Hach Chemical Company, Loveland, CO 80537. Available from: Hach Chemical Company, P.O. Box 389, Loveland, CO 80537.
- <sup>24</sup> Wershaw, R.L., et al, "Methods for Analysis of Organic Substances in Water," Techniques of Water-Resources Investigation of the U.S. Geological Survey, Book 5, Chapter A3, (1972 Revised 1987) p. 14. Available from: U.S. Geological Survey, 604 S. Pickett Street, Alexandria, VA 22304.
- <sup>25</sup> Nitrogen, Nitrite, Method 8507, Hach Chemical Company, POBox 389, Loveland, CO 80537. Available from: Hach Chemical Company, P.O. Box 389, Loveland, CO 80537.
- <sup>25m</sup> Nitrate-nitrite determinations by ion chromatography must be analyzed within 48 hours.
- <sup>26</sup> Just prior to distillation, adjust the sulfuric-acid-preserved sample to pH4 with 1 + 9 NaOH.
- <sup>27</sup> The approved method is cited in Standard Methods for the Examination of Water and Wastewater, 14th Edition. The colorimetric reaction is conducted at a pH of 10.0+/-0.2. The approved methods are given on pp 576-81 of the 14th Edition: Method 510A for distillation, Method 510B for the manual colorimetric procedure, or Method 510C for the manual spectrometric procedure. Available on inter-library loan.
- <sup>28</sup> R.F. Addison and R.G. Ackman, "Direct Determination of Elemental Phosphorus by Gas-Liquid Chromatography", Journal of Chromatography, Vol. 47, No. 3, pp. 421-426, 1970. Available in most public libraries. Back volumes of the Journal of Chromatography are available from: Elsevier/North-Holland, Inc., Journal Information Centre, 52 Vanderbilt Avenue, New York, NY 10164.
- <sup>29</sup> Approved methods for the analysis of silver in industrial wastewaters at concentrations of 1 mg/L and above are inadequate where silver exists as an inorganic halide. Silver halides such as the bromide and chloride are relatively insoluble in reagents such as nitric acid but are readily soluble in an aqueous buffer of sodium thiosulfate and sodium hydroxide to pH of 12. Therefore, for levels of silver above 1 mg/L, 20 mL of sample should be diluted to 100 mL by adding 40 mL each of 2 M Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub> and NaOH. Standards should be prepared in the same manner. For levels of silver below 1 mg/L the approved method is satisfactory.
- <sup>30</sup> The approved method is that cited in Standard Methods for the Examination of Water and Wastewater, 15th Edition. Available on interlibrary loan.
- <sup>31</sup> EPA Methods 335.2 and 335.3 require the NaOH absorber solution final concentration to be adjusted to 0.25 N before colorimetric determination of total cyanide.
- <sup>32</sup> Stevens, H.H., Ficke, J.F., and Smoot, G.F., "Water Temperature--Influential Factors, Field Measurement and Data Presentation", Techniques of Water-Resources Investigations of the U.S. Geological Survey, Book 1, Chapter D1, 1975. Available from: U.S. Geological Survey, 604 S. Pickett Street, Alexandria, VA 22304.
- <sup>33</sup> Zinc, Zincon Method, Method 8009, Hach Handbook of Water Analysis, 1979, pages 2-231 and 2-333, Hach Chemical Company, Loveland, CO 80537. Available from: Hach Chemical Company, P.O. Box 389, Loveland, CO 80537.
- <sup>34</sup> "Direct Current Plasma (DCP) Optical Emission Spectrometric Method for Trace Elemental Analysis of Water and Wastes, Method AES0029", 1986--Revised 1991, Thermo Jarrell Ash Corporation, 27 Forge Parkway, Franklin, MA 02038. Available from: the Thermo Jarrell Ash Corporation.
- <sup>35</sup> Precision and recovery statements for the atomic absorption direct aspiration and graphite furnace methods, and for the spectrophotometric SDDC method for arsenic are provided in Appendix D of this part titled, "Precision and Recovery Statements for Methods for Measuring Metals".
- <sup>36</sup> "Closed Vessel Microwave Digestion of Wastewater Samples for Determination of Metals", CEM Corporation, PO Box 200, Matthews, NC 28106-0200, April 16, 1992. Available from: the CEM Corporation.
- <sup>37</sup> When determining boron and silica, only plastic, PTFE, or quart zlaboratory ware may be used from start until completion of analysis.
- <sup>38</sup> Only use Trichlorotrifluorethane (1,1,2-trichloro-1,2,2-trifluoroethane; CFC-113) extraction solvent when determining Total Recoverable Oil and Grease (analogous to EPA Method 413.1). Only use n-hexane extraction solvent when determining Hexane Extractable Material (analogous to EPA Method 1664A). Use of other extraction solvents is strictly prohibited.
- <sup>39</sup> Nitrogen, Total Kjeldahl, Method PAI-DK01 (Block Digestion, Steam Distillation, Titrimetric Detection), revised 12/22/94, OI Analytical/ALPKEM, POBox 9010, College Station, TX 77842.
- <sup>40</sup> Nitrogen, Total Kjeldahl, Method PAI-DK02 (Block Digestion, Steam Distillation, Colorimetric Detection), revised 12/22/94, OI Analytical/ALPKEM, POBox 9010, College Station, TX 77842.

- <sup>41</sup> Nitrogen, Total Kjeldahl, Method PAI-DK03 (Block Digestion, Automated FIA Gas Diffusion), revised 12/22/94, OI Analytical/ALPKEM, PO Box 9010, College Station, TX 77842.
- <sup>42</sup> Method 1664, Revision A "n-Hexane Extractable Material (HEM; Oil and Grease) and Silica Gel Treated n-Hexane Extractable Material (SGT-HEM; Non-polar Material) by Extraction and Gravimetry" EPA-821-R-98-002, February 1999. Available at NTIS, PB-121949, U.S. Department of Commerce, 5285 Port Royal, Springfield, Virginia 22161.
- <sup>42m</sup> The full text of Method 1650 is given in Appendix A, "Methods 1650 and 1653", of 40 CFR Part 430. Available from: The Superintendent of Documents, U.S. Government Printing Office, Washington, D.C. 20402.
- <sup>43</sup> USEPA. 2002. Method 1631, Revision E, "Mercury in Water by Oxidation, Purge and Trap, and Cold Vapor Atomic Fluorescence Spectrometry". September 2002. Office of Water, U.S. Environmental Protection Agency (EPA-821-R-02-019). The application of clean techniques described in EPA's draft Method 1669: Sampling Ambient Water for Trace Metals at EPA Water Quality Criteria Levels (EPA-821-R-96-011) are recommended to preclude contamination at low-level, trace metal determinations. Available at NTIS, PB-121949, U.S. Depart ment of Commerce, 5285 Port Royal, Springfield, Virginia 22161.
- <sup>43m</sup> Quality control requirements for low level mercury are found in s. NR 106.145 (9) and (10), Wis. Adm. Code. Low-level mercury methods are performance-based so some method modifications are allowable, provided quality control requirements are met. If an atomic absorption detector is substituted for atomic fluorescence detector, the appropriate method citation is 245.1 (manual) or 245.2 (automated). If method 1631E is modified to eliminate the purge and trap step, the appropriate method citation is 245.7.
- <sup>44</sup> Available Cyanide, Method OIA-1677 (Available Cyanide by Flow Injection, Ligand Exchange, and Amperometry), ALPKEM, A Division of OI Analytical, PO Box 9010, College Station, TX 77842-9010.
- <sup>45</sup> "Methods of Analysis by the U.S. Geological Survey National Water Quality Laboratory-Determination of Ammonia Plus Organic Nitrogen by a Kjeldahl Digestion Method", Open File Report (OFR) 00-170. Available from: U.S. Geological Survey, 604 S. Pickett Street, Alexandria, VA 22304.
- <sup>46</sup> "Methods of Analysis by the U.S. Geological Survey National Water Quality Laboratory–Determination of Chromium in Water by Graphite Furnace Atomic Absorption Spectrophotometry", Open File Report (OFR) 93-449. Available from: U.S. Geological Survey, 604 S. Pickett Street, Alexandria, VA 22304.
- <sup>47</sup> "Methods of Analysis by the U.S. Geological Survey National Water Quality Laboratory–Determination of Molybdenum by Graphite Furnace Atomic Absorption Spectrophotometry", Open File Report (OFR) 97-198. Available from: U.S. Geological Survey, 604 S. Pickett Street, Alexandria, VA 22304.
- <sup>48</sup> "Methods of Analysis by the U.S. Geological Survey National Water Quality Laboratory-Determination of Total Phosphorus by Kjeldahl Digestion Method and an Automated Colorimetric Finish That Includes Dialysis" Open File Report (OFR) 92-146. Available from: U.S. Geological Survey, 604 S. Pickett Street, Alexandria, VA 22304.
- <sup>49</sup> "Methods of Analysis by the U.S. Geological Survey National Water Quality Laboratory-Determination of Arsenic and Selenium in Water and Sediment by Graphite Furnace-Atomic Absorption Spectrometry" Open File Report (OFR) 98-639. Available from: U.S. Geological Survey, 604 S. Pickett Street, Alexandria, VA 22304.
- <sup>50</sup> "Methods of Analysis by the U.S. Geological Survey National Water Quality Laboratory-Determination of Elements in Whole-water Digests Using Inductively Coupled Plasma-Optical Emission Spectrometry and Inductively Coupled Plasma-Mass Spectrometry", Open File Report (OFR) 98-165. Available from: U.S. Geological Survey, 604 S. Pickett Street, Alexandria, VA 22304.
- <sup>51</sup> "Methods of Analysis by the U.S. Geological Survey National Water Quality Laboratory-Determination of Inorganic and Organic Constituents in Water and Fluvial Sediment", Open File Report (OFR) 93-125. Available from: U.S. Geological Survey, 604 S. Pickett Street, Alexandria, VA 22304.

# SECTION 6. NR 219.04 TABLE BM is repealed and recreated to read:

#### TABLE BM

# METALS DIGESTION PROCEDURES

Analysis	SW-846 <sup>1</sup>	<b>EP</b> A <sup>2</sup>	EPA <sup>3</sup>
Dissolved Metals <sup>4</sup>	3005A, 3040A <sup>10</sup>		4.1.1
Suspended Metals <sup>5</sup>	3005A		4.1.2
Total Metals <sup>6</sup>	3010A, 3020A <sup>11</sup> , 3050B <sup>10</sup> , 3051A <sup>10</sup>		4.1.3
Total Recoverable Metals <sup>7</sup>	3005A	200.2	4.1.4
Acid Soluble Metals <sup>8</sup>		200.1 <sup>12</sup>	
Availible Metals <sup>9</sup>	3015A <sup>13</sup>		

SECTION 7. NR 219.04 TABLE BM Note 1 is repealed and recreated to read:

<sup>1</sup> "T est Methods for Evaluating Solid Waste, Physical/Chemical Methods", SW –846, EPA, Office of Solid Waste and Emergency Response, 401 M Street, S.W., Washington D.C. 20460, November, 1986 (third edition), including July 1992 (update I), September 1994 (up date II), January 1995 (update IIB), December 1996 (update III), January 1998 (update IVA), November 2000 (update IVB), August 2002 (update IIIB) updates. Available from: U.S. Government Printing Office (GPO), Superintendent of Documents, Washington, DC 20402, (202) 512-1800 (Publication Number: 955-001-00000-1). Also, available on-line at http://www.epa.gov/epaoswer/hazwaste/test/sw846.htm.

SECTION 8. NR 219.04 TABLE C is repealed and recreated to read:

Parameter <sup>1</sup>	EPA Me GC	thod Number <sup>2</sup> GC/MS	<sup>7</sup> Standard Methods <sup>2f</sup>	EPA Meth GC	od Number <sup>2m</sup> GC/MS	Other
I. VOLATILES		624 <sup>4</sup>		8021B	8260B	
A. Halogenated volatiles	601	1624B	6230 B[20 <sup>th</sup> ], 6210B[18 <sup>th</sup> ,19 <sup>th</sup> ], 6200 B[20 <sup>th</sup> ], 6200 C [20 <sup>th</sup> ]			
Bromodichloromethane			0200 C [20 ]			
Bromoform						
Bromomethane			(210 D			N
Carbon tetrachloride			not 6210 B, not 6200 B			Note 3, p.130
Chloroethane						
2-Chloroethylvinyl ether						
Chloroform						Note 3, p.130
Chloromethane						
Dibromochloromethane						
Dichlorodifluoromethane			not 6210 B, not 6200 B			
1,1-Dichloroethane						
1,2-Dichloroethane						
1,1-Dichloroethene						
trans-1,2-Dichloroethene						
1,2-Dichloropropane						
cis-1,3 Dichloropropene						
trans-1,3-Dichloropropene						
Methylene chloride	I		not 6210 B, not			Note 3, p.130

# TABLE C List of Approved Test Procedures for Non-Pesticide Organic Compounds in Wastewater

TABLE C
List of Approved Test Procedures for Non-Pesticide Organic Compounds in Wastewater

Parameter <sup>1</sup>	EPA Meth GC	od Number <sup>27</sup> GC/MS	Standard Methods <sup>2f</sup>	EPA Method GC	l Number <sup>2m</sup> GC/MS	Other
			6200 B			
1,1,2,2-Tetrachloroethane						Note 3, p.130
Tetrachloroethene						Note 3, p.130
1,1,1-Trichloroethane 1,1,2-Trichloroethane						Note 3, p.130
Trichloroethene						Note 5, p.150
Trichlorofluoromethane						
Vinylchloride						
B. Aromatic volatiles	602		6220 B [18 <sup>th</sup> ,19 <sup>th</sup> ],			
<b>D.</b> Afomatic volatiles	002		$6200 \text{ B} [20^{\text{th}}],$			
			6200 C [20 <sup>th</sup> ]			
Benzene		1624B	6210 B [18 <sup>th</sup> ,19 <sup>th</sup> ]			
Chlorobenzene	601	1624	6210 B [18 <sup>th</sup> ,19 <sup>th</sup> ], 6230 B [18 <sup>th</sup> ,19 <sup>th</sup> ]			Note 3, p.130
1,2-Dichlorobenzene	601,612	625 1625F	6230  B [18, 19] $6230 \text{ B} [18^{\text{th}}, 19^{\text{th}}],$			
1,2 Diemorobelizene	001,012	025,10251	6410 B			
			[18 <sup>th</sup> ,19 <sup>th</sup> ,20 <sup>th</sup> ]			
1,3-Dichlorobenzene	601,612	625,1625E	6230 B[18 <sup>th</sup> ,19 <sup>th</sup> ],			
			6410 B [18 <sup>th</sup> ,19 <sup>th</sup> ,20 <sup>th</sup> ]			
1,4-Dichlorobenzene	601,612	625 1625F	[18, 19, 20] 6230 B[18 <sup>th</sup> ,19 <sup>th</sup> ],			
	001,012	025,10251	6410 B			
			$[18^{\text{th}}, 19^{\text{th}}, 20^{\text{th}}]$			
Ethylbenzene		1624B	$6210 \mathrm{B} [18^{\mathrm{th}}, 19^{\mathrm{th}}]$			
Toluene		1624B	6210 B [18 <sup>th</sup> ,19 <sup>th</sup> ]			
C. O ther volatiles	603	1624B,624	1	8030A	8260B	
Acrolein						8315A <sup>2m</sup>
Acrylonitrile				8031		8316 <sup>2m</sup>
II. PHENOLS	604	625,1625B	6410 B, 6420 B	8041A	8270D	Note 9, p. 27
4 Chloro 2 methylphonol			[18 <sup>th</sup> , 19 <sup>th</sup> , 20 <sup>th</sup> ]			
4-Chloro-3-methylphenol 2-Chlorophenol						
2,4-Dichlorophenol						
2,4-Dimethlyphenol						
2,3-Dinitrophenol			Not 6410B			
2,4-Dinitrophenol						
2,6-Dinitrophenol			Not 6410B			
2-Methyl-4,6-dinitrophenol 2-Nitrophenol						
4-Nitrophenol						
Pentachlorophenol		1653 <sup>9s</sup>				Note 3, p.140
Phenol						
Trichlorosyringol	Not 604	1653 <sup>9s</sup> , not				
		625 or 1625B				
3,4,5-Trichlorocatechol	Not 604	$1623^{98}$ , not				
	1100 001	625 or				
		1625B				
3,4,6-Trichlorocatechol	Not 604	1653 <sup>9s</sup> , not				
		625 or 1625B				
3,4,5-Trichloroguaiacol	Not 604	1625B $1653^{9s}$ , not				
	101004	625 or				
		1625B				
3,4,6-Trichloroguaiacol	Not 604	1653 <sup>9s</sup> , not				
		625 or 1625B				
4,5,6-Trichloroguaiacol	Not 604	1623B $1653^{9s}$ , not				
.,.,. inclusional and a second	1.01.004	1055 ,1100	I	1		I

TABLE C
List of Approved Test Procedures for Non-Pesticide Organic Compounds in Wastewater

Parameter <sup>1</sup>	EPA Meth GC	od Number <sup>2,7</sup> GC/MS	Standard Methods <sup>2f</sup>	EPA Method GC	Number <sup>2m</sup> GC/MS	Other
		625 or				
		1625B				
2,4,5-Trichlorophenol	Not 604	1653 <sup>9s</sup> , not				
		625 or				
2.4.6 Tricklenenkenel		1625B 1653 <sup>9s</sup>				
2,4,6-Trichlorophenol Tetrachlorocatechol	Not 604	1653 <sup>9</sup> , not				
retractiorocatection	NOT 004	625 or				
		1625B				
Tetrachloroguaiacol	Not 604	1653 <sup>9s</sup> , not				
8		625 or				
		1625B				
2,3,4,6-Tetrachlorophenol	Not 604	1653 <sup>9s</sup> , not				
		625 or				
		1625B				
III. PHTHALATE ESTERS	606	625,1625B	6410 B	8061A	8270D	Note 9, p. 27
	000	023,10250	[18 <sup>th</sup> ,19 <sup>th</sup> ,20 <sup>th</sup> ]	0001A	0270D	1000 9, p. 27
Benzyl but yl phthalate						
Bis(2-ethylhexyl)phthalate						
Diet hyl phthalate						
Dimethyl phthalate						
Di-n-butyl phthalate						
Di-n-octylphthalate						
IV. NITROSAMINES	607	625,1625B	6410 B [18 <sup>th</sup> ,19 <sup>th</sup> ,20 <sup>th</sup> ]		8270D	Note 9, p. 27
N-Nitrosodimethylamine		Note 5	[18,19,20]			
N-Nitrosodi-n-propylamine						
N-Nitrosodiphenylamine		Note 5				
1 2						
V. POLYCHLORINATED BIPHENYLS	608	625	6410 B	8081B	8270D	Note 3, p.43
PCB-1016			[18 <sup>th</sup> ,19 <sup>th</sup> ,20 <sup>th</sup> ]			
PCB-1010 PCB-1221						
РСБ-1221 РСВ-1232						
PCB-1242						
PCB-1248						
PCB-1254						
PCB-1260						
VI. NITROAROMATICS & CYCLIC	609	625,1625B	6410 B	8091	8270D	Note 9, p. 27
KETONES			[18 <sup>th</sup> ,19 <sup>th</sup> ,20 <sup>th</sup> ]			
2,3-Dinitrotoluene						
2,4-Dinitrotoluene						
2,6-Dinitrotoluene						
Isophorone Nitrobenzene						Nata Ora
Nitrobenzene						Note 9m
VII. POLYNUCLEAR AROMATIC	610/FID	625.1625B	6410 B, 6440 B		8270D	Note 9, p. 27,
HYDROCARBONS	010/112	020,10202	$[18^{\text{th}}, 19^{\text{th}}, 20^{\text{th}}]$		02/02	Note 9m.
						610 <sup>2</sup> , 8310 <sup>2m</sup>
						8310 <sup>2m</sup>
Acenaphthene				1		
Acenaphthylene						
Anthracene						
Benzo(a)anthracene				1		
Benzo(a)pyrene						
Benzo(b)fluoranthene						
Benzo(g,h,i)perylene Benzo(k)fluoranthene						
DELIZOLK HUDOLZHUDEDE	1		1	1		1

TABLE C
List of Approved Test Procedures for Non-Pesticide Organic Compounds in Wastewater

Parameter <sup>1</sup>	EPA Method GC	l Number <sup>2,7</sup> GC/MS	Standard Methods <sup>2f</sup>	EPA Method GC	Number <sup>2m</sup> GC/MS	Other
Chrysene Dibenzo(a,h)anthracene Fluoranthene Fluorene Ideno (1,2-3-cd)pyrene Naphthalene Phenanthrene Pyrene				8021B		
VIII. HALOETHERS	611	625, 1625B	6410 B [18 <sup>th</sup> ,19 <sup>th</sup> ,20 <sup>th</sup> ]	8111	8270D	Note 9, p. 27
Bis(2-chloroethoxy) methane Bis(2-chloroethyl)ether 4-Bromophenylphenyl ether 4-Chlorophenylphenyl ether 2,2-Oxybis (2-chloropropane)			[10,19,20]			
IX. CHLORINATED HYDROCARBONS	612	625, 1625B	6410 B [18 <sup>th</sup> ,19 <sup>th</sup> ,20 <sup>th</sup> ]	8121	8270D 8260B	
Benzidine Benzyl chloride		Note 5			not 8260B	LC: 605 <sup>2</sup> Note 3, p.130; Note 6, p.S102
2-Chloronaphthalene					not 8260B	
3,3-Dichlorobenzidine Epichlorohydrin					not 8260B not 8270D	LC: 605 <sup>2</sup> Note 3, p.130; Note 6,
Hexachlorobenzene				8081B	not 8260B	p.S102 8410 <sup>2m</sup> , Note 9, p. 27
Hexachlorobutadiene				8021B		8410 <sup>2m</sup> , Note 9, p. 27
Hexachlorocyclopentadiene		Note 5		8081B	not 8260B	
Hexachloroethane	Not 612,616					8410 <sup>2m</sup> , Note 9, p. 27
1,2,4-Trichlorobenzene				8021B		Note 3, p. 130, Note 9, p. 27
X. POLYCHLORINATED DIBENZO-P- DIOXINS AND FURANS 1,2,3,4,6,7,8-Hept achlorodibenzo-p-dioxin 1,2,3,4,6,7,8-Hept achlorodibenzofuran 1,2,3,4,7,8-Hex achlorodibenzo-p-dioxin 1,2,3,4,7,8-Hex achlorodibenzo-p-dioxin 1,2,3,7,8-Hex achlorodibenzo-p-dioxin 1,2,3,7,8-Hex achlorodibenzofuran 1,2,3,6,7,8-Hex achlorodibenzofuran 1,2,3,7,8-Hex achlorodibenzofuran 1,2,3,7,8-Hex achlorodibenzofuran 1,2,3,7,8-Hex achlorodibenzofuran 2,3,4,6,7,8-Hex achlorodibenzofuran 0,0,4,7,8-Hex achlorodibenzofuran 1,2,3,7,8-Pent achlorodibenzo-p-dioxin 1,2,3,7,8-Pent achlorodibenzo-p-dioxin 1,2,3,7,8-Pent achlorodibenzo-p-dioxin 1,2,3,7,8-T etrachlorodibenzo-p-dioxin		1613 B <sup>9</sup>			8280B, 8290A	
2,3,7,8-T etrachlorodibenzo-p-dioxin 2,3,7,8-T etrachlorodibenzofuran		613 <sup>5m</sup>				

### SECTION 9. NR 219.04 TABLE C Notes are repealed and recreated to read:

#### TABLE C NOTES

- <sup>1</sup> All parameters are expressed in micrograms per liter ( $\mu$ g/L) except for Method 1613B in which the parameters are expressed in picograms per liter ( $\mu$ g/L).
- <sup>2</sup> The full text of Methods 601-613, 624, 625, 1624B, and 1625B, are given at Appendix A, ``Test Procedures for Analysis of Organic Pollutants", 'of 40 CFR Part 136. The full text of Method 1613B is incorporated by reference into this Part 136 and is Available from: the National Technical Information Services as stock number PB95-104774. The standardized test procedure to be used to determine the method detection limit (MDL) for these test procedures is given at Appendix B, "Definition and Procedure for the Determination of the Method Detection Limit", of 40 CFR Part 136. Available from: The Superintendent of Documents, U.S. Government Printing Office, Washington, D.C. 20402.
- <sup>2f</sup> "Standard Methods for the Examination of Water and Wastewater", Joint Editorial Board, American Public Health Association, American Water Works Association, and Water Pollution Control Federation, 20<sup>th</sup> Edition (1998), 19<sup>th</sup> Edition (1995), and 18<sup>th</sup> Edition, (1992). Available from: American Public Health Association, 1015 Fifteenth Street, N.W., Washington, D.C. 20005.
- <sup>2m</sup> "T est Methods for Evaluating Solid Waste, Physical/Chemical Methods", SW–846, EPA, Office of Solid Waste and Emergency Response, 401 M Street, S.W., Washington D.C. 20460, November, 1986 (third edition), including July 1992 (update I), September 1994 (up date II), January 1995 (update IIB), December 1996 (update III), January 1998 (update IVA), November 2000 (update IVB), August 2002 (update IIIB) updates. Available from: U.S. Government Printing Office (GPO), Superintendent of Documents, Washington, DC 20402, (202) 512-1800 (Publication Number: 955-001-00000-1). Also, available on-line at http://www.epa.gov/epaoswer/hazwaste/test/sw846.htm.
- <sup>3</sup> "Methods for Benzidine: Chlorinated Organic Compounds, Pentachlorophenol and Pesticides in Water and Wastewater", U.S. Environmental Protection Agency, September, 1978. Available from: ORD Publications, CERI, U.S. Environmental Protection Agency, 26 W. St. Claire, Cincinnati, Ohio 45268.
- <sup>4</sup> Method 624 may be extended to screen samples for Acrolein and Acrylonitrile. However, when they are known to be present, the preferred method for these two compounds is Method 603 or Method 1624B.
- <sup>5</sup> Method 625 may be extended to include benzidine, hexachlorocyclopentadiene, N-nitrosodimethylamine, and N-nitrosodiphenylamine. However, when they are known to be present, Methods 605, 607, and 612, or Method 1625B, are preferred methods for these compounds.
- <sup>6</sup> "Selected Analytical Methods Approved and Cited by the United States Environmental Protection Agency", Supplement to the Fifteenth Edition of Standard Methods for the Examination of Water and Wastewater (1981). Available from: American Public Health Association, 1015 Fifteenth Street, N.W., Washington, D.C. 20036.
- <sup>7</sup> Each Analyst must make an initial, one-time demonstration of their ability to generate acceptable precision and accuracy with Methods 601-603, 624, 625, 1624B, and 1625B (See Appendix A of this Part 136) in accordance with procedures each in Section 8.2 of each of these Methods. Additionally, each laboratory, on an on-going basis must spike and analyze 10% (5% for Methods 624 and 625 and 100% for Methods 1624B and 1625B) of all samples to monitor and evaluate laboratory data quality in accordance with Sections 8.3 and 8.4 of these Methods. When the recovery of any parameter falls outside the warning limits, the analytical results for that parameter in the unspiked sample are suspect and cannot be reported to demonstrate regulatory compliance. Note: These warning limits are promulgated as an "interim final action with a request for comments."
- <sup>8</sup> "Organochlorine Pesticides and PCBs in Wastewater Using Empore TM Disk" 3M Corporation, Revised 10/28/94. Method available from: 3M Corporation, 3M Center Building 220–9E–10, St. Paul, MN 55144–1000.
- <sup>9</sup> USGS Method 0-3116-87 from "Methods of Analysis by U.S. Geological Survey National Water Quality Laboratory Determination of Inorganic and Organic Constituents in Water and Fluvial Sediments", U.S. Geological Survey, Open File Report 93-125. Available from: U.S. Geological Survey, 604 S. Pickett Street, Alexandria, VA 22304.
- <sup>9f</sup> Method 1613: Tetra- through Octa- Chlorinated Dioxins and Furans by Isotope Dilution, HRGC/HRMS, Environmental Protection Agency, Federal Register, Office of Water, September 1994, Revision B, EPA 821-B-94-005. Available from: The Superintendent of Documents, US Government Printing Office, Washington, D.C. 20402.
- <sup>9m</sup> Method D4657-92, "Annual Book of Standards- Water and Environmental Technology", Section 11, Parts 11.01 and 11.02, American Society for Testing and Materials, 1993. Available from: the American Society for Testing and Materials, 1916 Race Street, Philadelphia, PA 19103.
- <sup>9s</sup> The full text of Method 1653 is given in Appendix A, "Methods 1650 and 1653", of 40 CFR Part 430. Available from: The Superintendent of Documents, U.S. Government Printing Office, Washington, D.C. 20402.

Parameter	Method EP	SW-	Standard Methods <sup>B</sup>	ASTM <sup>C</sup>	in Wastewater Other
1. Aldrin	GC 60		6630 B & C	D3086-90	Note 3, p.7; Note 4, p.27; Note 8.
	GC/MS 62	5 8270D	6410 B		
2. Ametryn	GC				Note 3, p.83; Note 6, p.868
3. Aminocarb	HPLC				Note 3m
4. Atraton	GC				Note 3, p.83; Note 6, p.S68
5. Atrazine	GC	8141B			Note 3. p.83; Note 6, p.S68
6. Azinphos methyl	GC	8141B			Note 3. p.25; Note 6, p.S51
	GC/MS	8270D			
7. Barban	HPLC				Note 3m
	GC/MS	8270D			
8. α-BHC	GC 60		6630 B & C	D3086-90	Note 3, p.7; Note 8.
	GC/MS 62				
9. β-BHC	GC 60		6630 C	D3086-90	Note 8.
	GC/MS 62			Dana c ac	
10. δ-BHC	GC 60			D3086-90	Note 8.
	GC/MS 62			Dana c ac	
11. γ-BHC (Lindane)	GC 60		6630 B & C	D3086-90	Note 3, p.7; Note 4, p.27; Note 8.
12 Conton	GC/MS 62	5 8270D		D2097 00	N-4-2 - 7
12. Captan	GC GC/MS	8270D	6630 B	D3086-90	Note 3. p.7.
12 (0.1)		8270D			No. 6 2 and
13. Carbaryl	HPLC	9270D			Note 3m
14 Carbonhanathian	GC/MS GC	8270D			Note 4 p 27: Note 8: Note 6 p 872
14. Carbophenothion	GC/MS	8141B 8270D			Note 4, p.27; Note 8; Note 6, p.S73.
15. Chlordane	GC 60		6630 B & C	D3086-90	Note 3, p.7; Note 4, p.27; Note 8
15. Chlordane	GC/MS 62			D3000-70	Note 5, p.7, Note 4, p.27, Note 8
16. Chloropropham	HPLC	02700	0410 D		Note 3m
17. 2,4-D	GC	81514	6640 B		Note 3, p.115; Note 4, p.40.
18. 4,4'-DDD	GC 60			D3086-90	Note 3, p.7; Note 4, p.27; Note 8.
101 1,1 222	GC/MS 62			20000 70	100001p11,10001,p127,100001
19.4,4'-DDE	GC 60			D3086-90	Note 3, p.7; Note 4, p.27; Note 8.
	GC/MS 62				
20. 4,4'-DDT	GC 60		6630 B & C	D3086-90	Note 3, p.7; Note 4, p.27; Note 8.
	GC/MS 62	5 8270D	6410 B		
21. Demeton-O	GC	8141B			Note 3, p.25; Note 6, p.851.
	GC/MS	8270D			
22. Demeton-S	GC	8141B			Note 3, p.25; Note 6, p.851.
	GC/MS	8270D			
23. Diazinon	GC	8141			Note 3, p.25; Note 4, p.27; Note 8; Note 6
					p.S51
24. Dicamba	GC	8151A			Note 3, p.115
25. Dichlofenthion	GC	8141			Note 4, p.27; Note8.; Note 6, p.S73
26. Dichloran	GC		6630 B & C	<b>D2004 00</b>	Note 3, p.7
27. Dicofol	GC	00015	6620 D 0 C	D3086-90	Note 2 = 7: Note 4 = 27 Note 9
28. Dieldrin	GC 60		6630 B & C		Note 3, p.7; Note 4, p.27; Note 8.
20 Diam (11)	GC/MS 62		6410 B		
29. Dioxathion	GC	8141B			Note 4, p.27; Note 8.; Note 6, p.S73
20 Dimiliator	GC/MS	8270D			Note 2 - 25: Note 6 - 551
30. Disulfoton	GC GC/MS	8141B 8270D			Note 3, p.25; Note 6, p.851
31. Diuron	GC/MS HPLC	8270D			Note 3m
31. Diuron 32. Endosulfan I	GC 60	8 8081B	6630 B & C	D3086-90	Note 3m Note 3, p.7; Note 4, p.27; Note 8.
2. Enuosuitait 1	GC/MS 62			D3000-70	11010 3, p. /, 11010 4, p. 27, 11010 6.
33. Endosulfan II	GC/MIS 62.			D3086-90	Note 3, p.7; Note 8
55. Enuosuitait II	GC/MS 62	-		D3000-90	1000 3, p.7, 1000 0
34. Endosulfan sulfate			6630 C		Note 8
J Enuosunan sunate	GC/MS 62		6410 B		1010 0
25 E 1				D3086-90	Note 3, p.7; Note 4, p.27; Note 8.
<ol><li>Endrin</li></ol>	GC 60			12.3000-90	

SECTION 10. NR 219.04 TABLE D is repealed and recreated to read:

Parameter	Method	EPA <sup>2,7</sup>	SW- 846 <sup>A</sup>	Standard Methods <sup>B</sup>	ASTM <sup>C</sup>	Other
36. Endrin aldehyde	GC	608	8081B		D3086-90	Note 8.
50. Endin udenyde	GC/MS	625	8270D	6410 B	<b>D</b> 5000 90	1010 0.
37. Ethion	GC/MIS	025	8141B	0410 D		Note 4, p.27; Note 8; Note 6, p.S73
57. Ethion	GC/MS		8141B 8270D			Note 4, p.27, Note 8, Note 0, p.375
20 F			8270D			
38. Fenuron	HPLC					Note 3, p.104; Note 6, p.S64
39. Fenuron-TCA	HPLC					Note 3m
40. Heptachlor	GC	608	8081B	6630 B & C	D3086-90	Note 3, p.7; Note 4, p.27; Note 8.
	GC/MS	625	8270D	6410 B		
41. Heptachlorepoxide	e GC	608	8081B	6630 B & C	D3086-90	Note 3, p.7; Note 4, p.27; Note 8; Note 6
						p.S73
	GC/MS	625	8270D	6410 B		
42. Isodrin	GC		8081B			Note 4, p.27; Note 8; Note 6, p.S73
	GC/MS		8270D			
43. Linuron	HPLC					Note 3m
44. Malathion	GC		8141B	6630 C		Note 3, p.25; Note 4, p.27; Note 8; Note 6
						p.S51
	GC/MS		8270D			I
45. Methiocarb	HPLC					Note10
46. Methoxychlor	GC		8081B	6630 B & C	D3086-90	Note 3, p.7; Note 4, p.27; Note 8.
40. Methoxychiol	GC/MS		8270D	0050 D & C	D3000-70	Note 5, p.7, Note 4, p.27, Note 6.
47. Mexacarbate	HPLC		8270D			Note 3m
47. Mexacal Date			0270D			Note 5III
49 Minan	GC/MS		8270D	((20 D € C		Nata 2 - 7: Nata 4 - 27
48. Mirex	GC		8081B	6630 B & C		Note 3, p.7; Note 4, p.27.
10.34	GC/MS		8270D			
49. Monuron	HPLC					Note 3m
50. Monuron-TCA	HPLC					Note 3m
51. Neburon	HPLC					Note 3m
52. Parathion methyl	GC		8141B	6630 C		Note 3, p.25; Note 4, p.27.
	GC/MS		8270D			
53. Parathion ethyl	GC		8141B	6630 C		Note 3, p.25
	GC/MS		8270D			
54. PCNB	GC		8081B	6630 B & C		Note 3, p.7
	GC/MS		8270 B			
55. Perthane	GC		8081B		D3086-90	Note 4, p. 27.
56. Prometon	GC					Note 3, p.83; Note 6, p.S68; Note 9.
57. Prometryn	GC					Note 3, p.83; Note 6, p.S68; Note 9.
58. Propazine	GC					Note 3, p.83; Note 6, p.S68; Note 9.
59. Propham	HPLC					Note 3m
60. Propoxur	HPLC					Note 3m
61. Secbumeton	HPLC					Note 3m
62. Siduron	HPLC					Note 3m
63. Simazine	GC		8141B			Note 3, p.83; Note 6, p.868; Note 9.
64. Strobane	GC			6630 B & C		
			00010	0030 B & C		Note 3, p.7
65. Swep	HPLC		01514	((10 D		Note 3m
66. 2,4,5-T	GC			6640 B		Note 3, p.115; Note 4, p.40.
67. 2,4,5-TP (Silvex)	GC		8151A	6640 B		Note 3, p.115; Note 4, p. 40.
68. Terbuthylazine	GC	105	005			Note 3, p.83; Note 6, p.868
69. Toxaphene	GC	608	8081B	6630 B & C	D3086-90	Note 3, p.7; Note 4, p.27; Note 8.
	GC/MS	625	8270D	6410 B		
70. Trifluralin	GC		8081B	6630 B		Note 3, p.7; Note 9.
	GC/MS		8270D			

 TABLE D

 List of Approved Test Procedures for Pesticides<sup>1</sup> in Wastewater

#### SECTION 11. NR 219.04 TABLE D Notes are repealed and recreated to read: TABLE D NOTES

<sup>A</sup> "Test Methods for Evaluating Solid Waste, Physical/Chemical Methods", SW–846, EPA, Office of Solid Waste and Emergency Response, 401 M Street, S.W., Washington D.C. 20460, November, 1986 (third edition), including July 1992 (update I), September 1994 (up date II), January 1995 (update IIB), December 1996 (update III), January 1998 (update IVA), November 2000 (update IVB), August 2002 (update IIIB) updates. Available from: U.S. Government Printing Office (GPO), Superintendent of Documents, Washington, DC 20402, (202) 512-1800 (Publication Number: 955-001-00000-1). Also, available on-line at http://www.epa.gov/epaoswer/hazwaste/test/sw846.htm.

- <sup>B</sup> "Standard Methods for the Examination of Water and Wastewater", Joint Editorial Board, American Public Health Association, American Water Works Association, and Water Pollution Control Federation, 20<sup>th</sup> Edition (1998), 19<sup>th</sup> Edition (1995), and 18<sup>th</sup> Edition, (1992). Available from: American Public Health Association, 1015 Fifteenth Street, N.W., Washington, D.C. 20005.
- <sup>C</sup> "Annual Book of Standards, Section 11.01 and 11.02, Water and Environmental Technology", American Society for Testing and Materials, 1994, 1996, and 1999. Available from: The American Society for Testing and Materials, 1916 Race Street, Philadelphia, PA 19103.
- <sup>1</sup> Pesticides are listed in this table by common name for the convenience of the reader. Additional pesticides may be found under Table C, where entries are listed by chemical name and type.
- <sup>2</sup> The full text of Methods 608 and 625 are given at Appendix A. "Test Procedures for Analysis of Organic Pollutants", of this Part 136. The standardized test procedure to be used to determine the method detection limit (MDL) for these test procedures is given at Appendix B, "Definition and Procedure for the Determination of the Method Detection Limit", of this Part 136. Available from: The Superintendent of Documents, U.S. Government Printing Office, Washington, D.C. 20402.
- <sup>3</sup> "Methods for Benzidine, Chlorinated Organic Compounds, Pentachlorophenol and Pesticides in Water and Wastewater,"U.S. Environmental Protection Agency, September 1978. This EPA publication includes thin-layer chromatography (TLC) methods. Available from: ORD Publications, CERI, U.S. Environmental Protection Agency, 26 W. St. Claire, Cincinnati, Ohio 45268.
- <sup>3m</sup> HPLC method 623 from "Methods for Nonconventional Pesticides Chemicals Analysis of Industrial and Municipal Wastewater", EPA 440/1-83/079-C, United States Environmental Protection Agency. Available from National Technical Information Service, 5258 Port Royal Road, Springfield, Virginia, 22161 (703) 487-4650.
- <sup>4</sup> "Methods for Analysis of Organic Substances in Water and Fluvial Sediments", Techniques of Water-Resources Investigations of the U.S. Geological Survey, Book 5, Chapter A3 (1987). Available from: U.S. Geological Survey, 604 S. Pickett Street, Alexandria, VA 22304.
- <sup>5</sup> The method may be extended to include [alpha]-BHC, [gamma]-BHC, endosulfan I, endosulfan II, and endrin. However, when they are known to exist, Method 608 is the preferred method.
- <sup>6</sup> "Selected Analytical Methods Approved and Cited by the United States Environmental Protection Agency". Supplement to the Fifteenth Edition of Standard Methods for the Examination of Water and Wastewater (1981). Available from: American Public Health Association, 1015 15th St., N.W., Washington, D.C. 20005.
- <sup>7</sup> Each analyst must make an initial, one-time, demonstration of their ability to generate acceptable precision and accuracy with Methods 608 and 625 (See Appendix A of this Part 136) in accordance with procedures given in Section 8.2 of each of these methods. Additionally, each laboratory, on an on- going basis, must spike and analyze 10% of all samples analyzed with Method 608 or 5% of all samples analyzed with Method 625 to monitor and evaluate laboratory dat a quality in accordance with Sections 8.3 and 8.4 of these methods. When the recovery of any parameter falls outside the warning limits, the analytical results for that parameter in the unspiked sample are suspect and cannot be reported to demonstrate regulatory compliance. These quality control requirements also apply to the Standard Methods, ASTM Methods, and other Methods cited. Note: These warning limits are promulgated as an "Interim final action with a request for comments." Available from: The Superintendent of Documents, U.S. Government Printing Office, Washington, D.C. 20402.
- <sup>8</sup> "Organochlorine Pesticides and PCBs in Wastewater Using Empore <sup>TM</sup> Disk", 3M Corporation, Revised 10/28/94. Method available from: 3M Corporation, 3M Center Building 220-9E-10, St. Paul, MN 55144-1000.
- <sup>9</sup> USGS Method 0-3106-93 from "Methods of Analysis by the U.S. Geological Survey National Water Quality Laboratory--Determination of Triazine and Other Nitrogen-containing Compounds by Gas Chromatography with Nitrogen Phosphorus Detectors" U.S. Geological Survey Open File Report 94-37. Available from: U.S. Geological Survey, 604 S. Pickett Street, Alexandria, VA 22304.

SECTION 12. NR 219.04 TABLE E is repealed and recreated to read:

	List of Approved Radiological lest liberules in wastewater					
				Standard		
Par	ameter and Units	Method	$EPA^{1}$	Methods <sup>2</sup>	ASTM <sup>3</sup>	$USGS^4$
1.	Alph-Total, pCi per liter	Proportional or Scintillation Counter	900.0	7110B	D1943-90	pp. 75 and 78 <sup>5</sup>
2.	Alpha-Counting error, pCi per liter	Proportional or Scintillation Counter	Appendix B	7110B	D1943-90	p. 79
3.	Beta-Total, pCi per liter	Proportional Counter	900.0	7110B	D1943-90	pp. 75 and 78 <sup>5</sup>
4.	Beta-Counting error, pCi	Proportional Counter	Appendix B	7110B	D1943-90	p. 79
5.	(a) Radium-Total	Proportional Counter	903.0	7500Ra B	D2460-90	
	(b) <sup>226</sup> Ra, pCi per liter	Scintillation Counter	903.1	7500 Ra C	CD3454-90	p. 81

# TABLE E List of Approved Radiological Test Procedures in Wastewater

#### SECTION 13. NR 219.04 TABLE E Notes are repealed and recreated to read: TABLE E NOTES:

- <sup>1</sup> "Prescribed Procedures for Measurement of Radioactivity in Drinking Water", EPA-600/-4-80-032, U.S. Environmental Protection Agency, August 1980.
- <sup>2</sup> "Standard Methods for the Examination of Water and Wastewater", 18th, 19th, or 20th Edition, Joint Editorial Board, American Public Health Association, American Water Works Association, and Water Pollution Control Federation, 1015 Fifteenth Street, N.W., Washington, D.C. 20005, 1992, 1995, or 1998. Available from: American Public Health Association, 1015 Fifteenth Street, N.W., Washington, D.C. 20005.
- <sup>3</sup> "1991 Annual Book of Standards, Water" Section 11, American Society for Testing and Materials, 1980. Available from: American Society for Testing and Materials, 1916 Race Street, Philadelphia, PA 19103.
- <sup>4</sup> "Selected Methods of the U.S. Geological Survey of Analysis of Wastewaters," U.S. Geological Survey, Open-File Report 76-177 (1976)
- <sup>5</sup> The method found on p. 75 measures only the dissolved portion while the method on p. 78 measures only the suspended portion. Therefore, the two results must be added to obtain the "total".

Table EM           Approved Analytical Methods For Sludge						
Parameter	Sample Preparation	Method	Method Number			
Metals <sup>1</sup>						
Arsenic	3050B	Inductively Coupled Plasma Emission	6010C			
Arsenic	7061A	Gaseous Hydride <sup>2</sup>	7061A			
Arsenic	3050B	Graphite Furnace	7010			
Beryllium	3050B	Inductively Coupled Plasma Emission	6010C			
Beryllium	3050B	Flame Atomic Absorption	7000B			
Beryllium	3050B	Graphite Furnace	7010			
Cadmium	3050B	Inductively Coupled Plasma Emission	6010C			
Cadmium	3050B	Flame Atomic Absorption	7000B			
Cadmium	3050B	Graphite Furnace	7010			
Chromium	3050B	Inductively Coupled Plasma Emission	6010C			
Chromium	3050B	Flame Atomic Absorption	7000B			
Chromium	3050B	Graphite Furnace	7010			
Copper	3050B	Inductively Coupled Plasma Emission	6010C			
Copper	3050B	Flame Atomic Absorption	7000B			
Lead	3050B	Inductively Coupled Plasma Emission	6010C			
Lead	3050B	Flame Atomic Absorption	7000B			
Lead	3050B	Graphite Furnace <sup>3</sup>	7010			
Mercury	7471B	Cold Vapor	7471B			
Molybdenum	3050B	Inductively Coupled Plasma Emission	6010C			
Molybdenum	3050B	Graphite Furnace	7010			
Nickel	3050B	Inductively Coupled Plasma Emission	6010C			
Nickel	3050B	Flame Atomic Absorption	7000B			
Selenium	3050B	Inductively Coupled Plasma Emission	6010C			
Selenium	7741A	Gaseous Hydride <sup>2</sup>	7741A			
Selenium	3050B	Graphite Furnace	7010			
Zinc	3050B	Inductively Coupled Plasma Emission	6010C			

SECTION 14. NR 219.04 TABLE EM is repealed and recreated to read:

Table EM							
Approved Analytical Methods For Sludge							
Parameter	Method	Method Number					
Zinc	3050B	Flame Atomic Absorption	7000B				
Organic <sup>1</sup>							
PCB (Aroclor or Congeners)	3540C or 3545A	Gas Chromatography	8082A <sup>11</sup>				
PCB (Congeners)	1668A	Gas Chromatography/Mass Spectrometry	1668A <sup>10,12</sup>				
Biological							
Enteric viruses	NA	Centrifuge Concentration	D 4994-89 <sup>4</sup> Appendix H <sup>8</sup>				
Fecal coliform	NA	Most Probable Number Membrane Filter	9221 E or 9222 $D^5$ or Appendix $F^8$				
Helminth ova	NA	Density Gradient Flotation	<sup>6</sup> or Appendix I <sup>8</sup>				
Specific Oxygen Uptake Rate	NA	Respirometer	2710 B <sup>5</sup> or Appendix D.2. <sup>8</sup>				
Salmonella	NA	Most Probable Number Selective Media Culture	9260 D.1 <sup>5</sup> or Appendix $G^8$				
Physical							
Solids	NA	Gravimetric	2540 G <sup>5</sup>				
Percent Volatiles Solids Reduction	NA	Calculation	Appendix D.1. & 3. <sup>8</sup>				

SECTION 15. NR 219.04 TABLE EM Notes are repealed and recreated to read:

#### TABLE EM NO TES

- <sup>1</sup> "T est Methods for Evaluating Solid Waste", Third Edition, SW–846, Office of Solid Waste and Emergency Response, Environmental Protection Agency, November 1986, including December 1987, July 1992, September 1992, August 1993, September 1994, January 1995, December 1996, January 1998, and November 2000 updates, Washington, DC 20460. Available from: The Superintendent of Documents, U.S. Government Printing Office, Room 190, Federal Building, P.O. Box 371954, Pittsburgh, PA 15250–7954, (202) 783–3238. Available online at http://www.epa.gov/epaoswer/hazwaste/test/sw846.htm.
- <sup>2</sup> High levels of chromium, copper, mercury, silver, cobalt, or molybdenum may interfere with the analysis. Consult Method 3114, of "Standard Method for the Examination of Water and Wastewater", 18<sup>th</sup>, 19<sup>th</sup>, or 20<sup>th</sup> edition, for more information.
- <sup>3</sup> Concentrations of lead in municipal sludge may exceed the working range of Graphite Furnace.
- <sup>4</sup> "1993 Annual Book of ASTM Standards, Section 11.02, Water and Environmental Technology", American Society for Testing and Materials, 1993, 1916 Race Street, Philadelphia, PA 19103. Available from: the American Society for Testing and Materials, 1916 Race Street, Philadelphia, PA 19103.
- <sup>5</sup> "Standard Methods for the Examination of Water and Wastewater", 18<sup>th</sup>, 19<sup>th</sup>, or 20<sup>th</sup> ed., American Public Health Association, 1015 Fifteenth Street NW, Washington D.C. 20005, 1992, 1995, or 1998. Available from: American Public Health Association, 1015 Fifteenth Street, N.W., Washington, D.C. 20005.
- <sup>6</sup> "Occurrence of Pathogens in Distribution and Marketing Municipal Sludges", EPA 600/1–87–014, Environmental Protection Agency, 1987. Available from: the National Technical Information Service, order # PB 88–154273/AS, 5285 Port Royal Road, Springfield, Virginia 22161,(703)487–4650.
- <sup>8</sup> "Environmental Regulations and Technology Control of Pathogens and Vectors Attraction in Sewage Sludge", EPA–625/R–92/013, Revised October 1999, Environmental Protection Agency, Cincinnati, OH, 1999. Available from: the National Technical Information Service, 5285 Port Royal Road, Springfield, Virginia 22161, (703) 487–4650.
- <sup>9</sup> If an alternative digestion procedure is specified in the analytical method, the digestion in the method shall be used. In all cases, consult the analytical method for special requirements and cautions. SW-846 method 3051A is an acceptable alternate digestion procedure to SW-846 method 3050B.

- <sup>10</sup> EPA Method 1668 may be used to test for all PCB congeners. If this method is employed, all PCB congeners shall be delineated. Nondetects shall be treated as zero. The values that are between the limit of detections and the limit of quantitation shall be used when calculating the total value of all congeners. All results shall be added together and the total PCB concentration by dry weight reported. It is recognized that a number of the congeners will co-elute with others, so there will not be 209 results to sum.
- EPA Method 8082A shall be used for PCB-Aroclor analysis and may be used for congener specific analysis as well. If congener specific analysis is performed using Method 8082A, the list of congeners tested shall include at least congener numbers 5, 18, 31, 44, 52, 66, 87, 101, 110, 138, 141, 151, 153, 170, 180, 183, 187, and 206 plus any other additional congeners which might be reasonably expected to occur in the particular sample. For either type of analysis, the sample shall be extracted using Soxhlet extraction Method 3540C or Pressurized Fluid Extraction Method 3545A. If Aroclor analysis is performed using Method 8082A, clean up steps of the extract shall be performed as necessary to remove interference and achieve as close to a limit of detection of 0.11 mg/kg as possible. If congener specific analysis is done using Method 8082A, clean up steps of the extract shall be performed as necessary to remove interference and achieve as close to a limit of detection of 0.01 mg/kg as possible. If congener specific analysis is done using Method 8082A, clean up steps of the extract shall be performed as necessary to remove interference and to achieve as close to a limit of detection of 0.003 mg/kg as possible for each congener. If the aforementioned limits of detection cannot be achieved after using the appropriate clean up techniques, a reporting limit that is achievable for the aroclors or each congener for sample shall be determined. This report limit should be reported and qualified indicating the presence of an interference. The lab conducting the analysis shall perform as many the following methods as necessary to remove interference:

3620C - Florisil 3640A – Gel Permeation 3630C - Silica Gel 3611B - Alumina 3660B - Sulfur Clean Up 3665A – Sulfuric Acid Clean Up

<sup>12</sup> "Method 1668A, Revision A: Chlorinated Biphenyl Congeners in Water, Soil, Sediment, and Tissue by HRGC/HRMS", EPA-821-R-00-002, Environmental Protection Agency, Office of Water, Washington, D.C., December 1999. Available from: the National Technical Information Service, 5285 Port Royal Road, Springfield, Virginia 22161, (703)487–4650.

Table ES

SECTION 16. NR 219.04 TABLE ES is created to read:

List of Approved Methods for Pharmaceutical Pollutants <sup>1a</sup>				
Pharmaceuticals pollutants	CAS registry No.	Analytical method number <sup>1m</sup>		
acetonitrile	75-05-8	1666/1671/D3371/D3695.		
n-amyl acetate	628-63-7	1666/D3695.		
n-amyl alcohol	71-41-0	1666/D3695		
benzene	71-43-2	D4763/D3695/502.2/524.2.		
n-butyl-acetate	123-86-4	1666/D3695.		
tert-butyl alcohol	75-65-0	1666		
chlorobenzene	108-90-7	502.2/524.2.		
chloroform	67-66-3	502.2/524.2/551.		
o-dichlorobenzene	95-50-1	1625C/502.2/524.2.		
1,2-dichloroethane	107-06-2	D3695/502.2/524.2.		
diethylamine	109-89-7	1666/1671.		
dimethyl sulfoxide	67-68-5	1666/1671.		
ethanol	64-17-5	1666/1671/D3695.		
ethyl acetate	141-78-6	1666/D3695.		
n-heptane	142-82-5	1666/D3695.		
n-hexane	110-54-3	1666/D3695.		
isobutyraldehyde	78-84-2	1666/1667.		
isopropanol	67-63-0	1666/D3695.		
isopropyl acetate	108-21-4	1666/D3695.		
isopropyl ether	108-20-3	1666/D3695.		
methanol	67-56-1	1666/1671/D3695.		
Methyl Cellosolve [Delta]	109-86-4	1666/1671		
methylene chloride	75-09-2	502.2/524.2		
methyl formate	107-31-3	1666		
4-methyl-2-pentanone (MIBK)	108-10-1	1624C/1666/D3695/D4763/524.2.		
phenol	108-95-2	D4763.		

List of Approved Methods for Pharmaceutical Pollutants <sup>1a</sup>			
Pharmaceuticals pollutants	CAS registry No.	Analytical method number <sup>1m</sup>	
n-propanol	71-23-8	1666/1671/D3695.	
2-propanone (acetone)	67-64-1	D3695/D4763/524.2.	
tetrahydrofuran	109-99-9	1666/524.2.	
toluene	108-88-3	D3695/D4763/502.2/524.2.	
triethlyamine	121-44-8	1666/1671.	
xylenes	(Note 1)	1624C/1666.	

 Table ES

 List of Approved Methods for Pharmaceutical Pollutants <sup>1a</sup>

SECTION 17. NR 219.04 TABLE ES Notes are created to read:

### TABLE ES NOTES

- 1 1624C: m-xylene 108-38-3, o,p-xylene E-14095 (Not a CAS number; this is the number provided in the Environmental Monitoring Methods Index (EMMI) database.); 1666: m,p-xylene 136777-61-2, o-xylene 95-47-6.
- <sup>1a</sup> Test methods listed in Table C maybe use for the parameters listed in this table.
- <sup>1m</sup> EPA Methods 1666, 1667, and 1671 listed in the table above are published in the compendium titled Analytical Methods for the Determination of Pollutants in Pharmaceutical Manufacturing Industry Wastewaters (EPA 821-B-98-016). EPA Methods 502.2 and 524.2 have been incorporated by reference into 40 CFR 141.24 and are in Methods for the Determination of Organic Compounds in Drinking Water, EPA-600/4-88-039, December 1988, Revised, July 1991, and Methods for the Determination of Organic Compounds in Drinking Water-Supplement II, EPA-600/R-92-129, August 1992, respectively. These EPA test method compendia are available from the National Technical Information Service, NTIS PB91-231480 and PB92-207703, U.S. Department of Commerce, 5285 Port Royal Road, Springfield, Virginia 22161. The toll-free number is 800-553-6847. AST M test methods D3371, D3695, and D4763 are available from the American Society for Testing and Materials, 100 Barr Harbor Drive, West Conshohocken, PA 19428-2959.

SECTION 18. NR 219.04 TABLE F is repealed and recreated to read:

Param	eter No./Name	Container <sup>1</sup>	Preservation <sup>2,3</sup>	Maximumholding time <sup>4</sup>
TABL	E A - Bacterial Tests:			
1-7.	Bacteria	P,G	Cool, <10°C, 0.0008%, Na <sub>2</sub> S <sub>2</sub> O <sub>3</sub> <sup>5</sup>	6 hours
8-9.	Protozoa	LDPE	0-8°C	96 hours <sup>17a</sup>
9-12.	Acute & chronic toxicity	P,G	$Cool, \leq 6^{\circ}C^{16}$	36 hours
TABL	E B - Inorganic Tests:			
1.	Acidity	P,G	Cool, ≤6°C	14 days
2.	Alkalinity	P,G	Cool, ≤6°C	14 days
4.	Ammonia	P,G	Cool, ≤6°C, H₂SO₄ to pH<2	28 days
9.	Biochemical oxygen demand	l P,G	Cool, ≤6°C	48 hours
11.	Bromide	P,G	None required	28 days
14.	Biochemical oxygen demand carbonaceous	l, P,G	Cool, ≤6°C	48 hours
15.	Chemical oxygen demand	P,G	Cool, $\leq$ 6°C, H <sub>2</sub> SO <sub>4</sub> to pH<2	28 days
16.	Chloride	P,G	None required	28 days
17.	Chlorine, total residual	P,G	None required	Analyze immediately
21.	Color	P,G	Cool, ≤6°C	48 hours
23-24.	Cyanide: total, amenable to chlorination, and available	P,G	Cool, ≤6°C, NaOH to pH>12, 0.6g as	corbic acid <sup>5</sup> 14 days <sup>6</sup>
25.	Fluoride	Р	None required	28 days

TABLE F REQUIRED CONTAINERS, PRESERVATION TECHNIQUES, AND HOLDING TIMES FOR WASTEWATER

Param	eter No./Name	Container <sup>1</sup>	Preservation <sup>2,3</sup>	Maximumholding time <sup>4</sup>
27.	Hardness	P,G	HNO <sub>3</sub> to pH<2, H <sub>2</sub> SO <sub>4</sub> to pH<2	6 months
28.	Hydrogen ion (pH)	P,G	None required	Analyze immediately
31,43.	Kjeldahl and organic nitroger	nP,G	Cool, $\leq$ 6°C, H <sub>2</sub> SO <sub>4</sub> to pH<2	28 days
38.	Nitrate	P,G	Cool, ≤6°C	48 hours
39.	Nitrate-nitrite	P,G	Cool, $\leq$ 6°C, H <sub>2</sub> SO <sub>4</sub> to pH<2	28 days
40.	Nitrite	P,G	Cool, ≤6°C	48 hours
41.	Oil and grease	G	Cool, ≤6°C, HCl or H₂SO₄ to pH<2	28 days
42.	Organic carbon	G	Cool, $\leq 6^{\circ}$ C, HCl or H <sub>2</sub> SO <sub>4</sub> or H <sub>3</sub> PO <sub>4</sub> to pH<2	28 days
44.	Orthophosphate	P,G	Filter immediately, Cool, ≤6°C	48 hours
46.	Oxygen, Dissolved Probe	G Bottle and top	None required	Analyze immediately
47.	Winkler	G Bottle and top	Fix on site and store in dark	8 hours
48.	Phenols	Gonly	Cool, ≤6°C, H₂SO₄ to pH<2	28 days
49.	Phosphorus (elemental)	G	Cool, ≤6°C	48 hours
50.	Phosphorus, total	P,G	Cool, ≤6°C, H₂SO₄ to pH<2	28 days
53.	Residue, total	P,G	Cool, ≤6°C	7 days
54.	Residue, Filterable	P,G	Cool, ≤6°C	7 days
55.	Residue, Nonfilterable (TSS)	P,G	Cool, ≤6°C	7 days
56.	Residue, Settleable	P,G	Cool, ≤6°C	48 hours
57.	Residue, Volatile	P,G	Cool, ≤6°C	7 days
61.	Silica	P, or Quartz	Cool, ≤6°C	28 days
64.	Specific conductance	P,G	Cool, ≤6°C	28 days
65.	Sulfate	P,G	Cool, ≤6°C	28 days
66.	Sulfide	P,G	Cool, ≤6°C, add zinc acetate plus NaOH to pH >9	7 days
67.	Sulfite	P,G	None required	Analyze immediately
68.	Surfactants	P,G	Cool, ≤6°C	48 hours
69.	Temperature	P,G	None required	Analyze immediately
73.	Turbidity	P,G	Cool, ≤6°C	48 hours
TABLE	$E B - Metals^7$ :			
10.	Boron	P, or Quartz	HNO <sub>3</sub> to pH<2	6 months
18.	Chromium VI <sup>7</sup>	P,G	Cool, ≤6°C	24 hours
35.	Mercury <sup>17b</sup>	P,G	HNO <sub>3</sub> to pH<2	28 days
35m.	Mercury (II) & Organomercurials	Amber G	Cool, ≤6°C	7 days
71.	Tin	Р	HCl or HNO <sub>3</sub> to pH<2	6 months
3, 5-8,	Metals:	P,G	HNO <sub>3</sub> to pH<2	6 months
10, 12, 13, 19, 20, 22, 26, 29, 30, 32- 34, 36, 37, 45, 47, 51, 52, 58- 60, 62, 63, 70- 72, 74, 75.	(except Cr VI, Sn, Hg <sup>7</sup> , & B)			

	TABLE F	
DEO LIDED CONTAINEDS	DDESEDVATION TECHNIQUES	AND HOLDING TIMES FOR WASTEWATED

Param	eter No./Name	Container <sup>1</sup>	HNIQUES, AND HOLDING TIMES FOR WA Preservation <sup>2,3</sup>	Maximumholding time <sup>4</sup>
[ABL]	E C - Organic Tests <sup>8</sup> :			
[A.	Purgeable halocarbons	G, Teflon-lined septum	Cool, $\leq 6^{\circ}$ C, 0.008% Na <sub>2</sub> S <sub>2</sub> O <sub>3</sub> <sup>5</sup>	14 days
[ <b>B</b> .	Purgeable aromatics	G, Teflon-lined septum	Cool, $\leq$ 6°C, 0.008% Na <sub>2</sub> S <sub>2</sub> O <sub>3</sub> <sup>5</sup> , HCl to pH<2	14 days
IC.	Acrolein and acrylonitrile	G, Teflon-lined septum	Cool, ≤6°C, 0.008% Na <sub>2</sub> S <sub>2</sub> O <sub>3</sub> <sup>5</sup> Adjust pH to 4-5 <sup>10</sup>	14 days
II.	Phenols <sup>11</sup>	G, Teflon-lined cap	Cool, ≤6°C, 0.008% Na <sub>2</sub> S <sub>2</sub> O <sub>3</sub> <sup>5</sup>	7 days until extraction; 40 days after extraction
X.	Benzidines (Benzidine and 3,3- Dichlorobenzidine) <sup>11</sup>	G, Teflon-lined cap	Cool, ≤6°C, 0.008% Na <sub>2</sub> S <sub>2</sub> O <sub>3</sub> <sup>5</sup>	7 days after extraction <sup>13</sup>
III.	Phthlate esters <sup>11</sup>	G, Teflon-lined cap	Cool, ≤6°C	7 days until extraction; 40 days after extraction
IV.	Nitrosamines <sup>11, 14</sup>	G, Teflon-lined cap	Cool, ≤6°C, store in dark, 0.008% Na <sub>2</sub> S <sub>2</sub> O <sub>3</sub> <sup>5</sup>	7 days until extraction; 40 days after extraction
V.	PCBs <sup>11</sup>	G, Teflon-lined cap	Cool, ≤6°C	7 days until extraction; 40 days after extraction
VI.	Nitroaromatics, cyclic ketone and isophorone <sup>11</sup>	s G, Teflon-lined cap	Cool, ≤6°C, store in dark, 0.008% Na <sub>2</sub> S <sub>2</sub> O <sub>3</sub> <sup>5</sup>	7 days until extraction; 40 days after extraction
VII.	Polynuclear aromatic hydrocarbons <sup>11</sup>	G, Teflon-lined cap	Cool, ≤6°C, store in dark, 0.008% Na <sub>2</sub> S <sub>2</sub> O <sub>3</sub> <sup>5</sup>	7 days until extraction; 40 days after extraction
VIII.	Haloethers <sup>11</sup>	G, Teflon-lined cap	Cool, ≤6°C, 0.008% Na <sub>2</sub> S <sub>2</sub> O <sub>3</sub> <sup>5</sup>	7 days until extraction; 40 days after extraction
X.	Chlorinated hydrocarbons <sup>11</sup>	G, Teflon-lined cap	Cool, ≤6°C	7 days until extraction; 40 days after extraction
X.	Chorinated Dioxans and Furans	G, Teflon-lined cap	$Cool, \le \!\!6^{\circ}\!C, pH\!<\!\!9, 0.008\% Na_2S_2O_3^{5}$	1 year
[ABL]	E E - Pesticide Tests:			
1-70.	Pesticides <sup>11</sup>	G, Teflon-lined cap	Cool, ≤6°C, pH 5-9 <sup>15</sup>	7 days until extraction; 40 days after extraction
[ABL]	E F - Radiological Tests:			
1-5.	Alpha, beta, and radium	P,G	HNO <sub>3</sub> to pH<2	6 months

TABLE F
REQUIRED CONTAINERS, PRESERVATION TECHNIQUES, AND HOLDING TIMES FOR WASTEWATER

SECTION 19. NR 219.04 TABLE F Notes are repealed and recreated to read:

### TABLE F NO TES

Polyethylene (P), or glass (G). For bacteria, plastic sample containers must be made of sterilizable materials (polypropylene [PP] or other autoclavable plastic). For protozoa, plastic sample containers must be made of low-density polyethylene (LDPE). For samples collected for trace-level mercury see note 17b.

All samples requiring preservation at ≤6°C must be cooled immediately after collection, but not frozen. The temperature of the samples shall be documented upon receipt at the laboratory. If the samples are shipped in crushed or cube ice (not "blue ice" packs) and solid ice is still present in the cooler, the lab may simply report the samples as "received on ice". If the ice has melted, the lab must report the either the temperature of the meltwater or of a temperature blank. A temperature blank is defined as an aliquot of deionized water, in an appropriate sample container, which is transported along with the samples. If sampling teams use "blue ice" packs, it is necessary to pre-chill all sample containers to at least 6 degrees celsius with ice or refrigeration prior to shipping. Since shipping simply with "blue ice" packs does not insure that samples are maintained at the appropriate temperatures, the sample collector must submit a temperature blank when using these

ice packs for shipping. For composite chemical samples each aliquot should be preserved at the time of collection. When use of an automated sampler makes it impossible to preserve each aliquot, then chemical samples may be preserved by maintaining at  $\leq 6^{\circ}$ C until compositing and sample splitting are completed. For low-level mercury see note 17b.

- <sup>3</sup> When any sample is to be shipped by common carrier or sent through the United States mail, it must comply with the Department of Transportation Hazardous Materials Regulations (49 CFR Part 172). The person offering such material for transportation is responsible for ensuring such compliance. For the preservation requirements of Table F, the Office of Hazardous Materials, Materials Transportation Bureau, Department of Transportation has determined that the Hazardous Materials Regulations do not apply to the following materials: Hydrochloric acid (HCl) in water solutions at concentrations of 0.04% by weight or less (pH about 1.96 or greater); Nitric acid (HNO<sub>3</sub>) in water solutions at concentrations of 0.15% by weight or less (pH about 1.62 or greater); Sulfuric acid (H<sub>2</sub>SO<sub>4</sub>) in water solutions at concentrations of 0.35% by weight or less (pH about 1.15 or greater); and Sodium hydroxide (NaOH) in water solutions at concentrations of 0.080% by weight or less (pH about 12.30 or less).
- <sup>4</sup> Samples should be analyzed as soon as possible after collection. The times listed are the maximum times that samples may be held before analysis and still be considered valid. For other composite samples, the holding time commences immediately after the samples are removed from the composite sampler. The time the sample spends in the sampler during collection does not count towards the maximum holding time. Samples for biological or chemical analysis may be held for longer periods than specified in this table only if the permittee or monitoring laboratory, has data on file to show that the specific types of samples under study are stable for the longer time, and has received a variance from the Regional Administrator (s. NR 219.05). Some samples may not be stable for the maximum time period given in the table. A permittee or monitoring laboratory is obligated to hold the sample for a shorter time if knowledge exists to show that this is necessary to maintain sample stability. See sec. 40 CFR 136.3(e) for details. The term "analyze immediately" usually means within 15 minutes or less of sample collection.
- <sup>5</sup> Should only be used in the presence of residual chlorine.
- Immediately after collection, preserve the sample using any or all of the following techniques, as necessary, followed by adjustment of the sample pH to 12 by addition of sodium hydroxide and refrigeration as specified: (1) Sulfide: The maximum holding time for an untreated sample is 24 hours when sulfide is present. Optionally, the sample may be treated and the maximum holding time extended to 14 days. Generally, the laboratory should test the sample with lead acetate test paper to determine the presence or absence of sulfide ion. However, for cyanide methods using amperometric detection systems (e.g., OIA-1677 for available cyanide), sulfide levels below those detectable with lead acet at e paper (approximately 5 ppm) may produce a false positive signal for cyanide. If there is reason to suspect sulfide levels below the detectable level of lead acetate paper when using an amperometric method, test the sample using a more sensitive sulfide method to determine if the treatment (described below) is required. If sulfide ion is present, treat the sample immediately (within 15 minutes of collection) with sufficient solid lead carbonate to remove sulfide (as evidenced by a lead acetate test paper), and immediately filter into another sample bottle to remove precipitated lead sulfide. If sulfide ion is suspected to be present, but its presence is not detected by the lead acetate paper test, two samples must be collected. One is treated for the presence of sulfide and immediately filtered, while the second is not treated for sulfide. Analyze both samples and report the lower of the two results. (2) Sulfide and particulate matter: If the sample contains sulfide and particulate matter that would be removed by filtration, filter the sample prior to treatment with lead carbonate to assure that cyanides associated with the particulate matter are included in the measurement. Save the particulate matter and treat the filtrate using the sulfide removal procedure above. Combine and homogenize the collected particulate and treated filtrate prior to shipment to the laboratory for analysis.
- <sup>7</sup> Samples should be filtered immediately on-site before adding preservative for dissolved metals, except for samples collected for trace-level mercury (see note 17b).
- <sup>8</sup> Guidance applies to samples to be analyzed by GC, LC, or GC/MS for specific compounds.
- <sup>9</sup> Samples receiving no pH adjustment must be analyzed within 7 days of sampling.
- <sup>10</sup> The pH adjustment is not required if acrolein will not be measured. Samples for acrolein receiving no pH adjustment must be analyzed within 3 days of sampling.
- <sup>11</sup> When the extractable analytes of concern fall within a single chemical category, the specified preservation and maximum holding times should be observed for optimum safeguard of sample integrity. When the analytes of concern fall within 2 or more chemical categories, the sample may be preserved by cooling to  $\leq 6^{\circ}$ C, reducing residual chlorine with 0.008% sodium thiosulfate, storing in the dark, and adjusting the pH to 6-9; samples preserved in this manner may be held for 7 days before extraction and for 40 days after extraction. Exceptions to this optional preservation and holding time procedure are noted in footnote 5 (re the requirement for thiosulfate reduction of residual chlorine), and footnotes 12, 13 (re the analysis of benzidine).
- <sup>12</sup> If 1,2-diphenylhydrazine is likely to be present, adjust the pH of the sample to  $4.0 \pm 0.2$  to prevent rearrangement to benzidine.
- <sup>13</sup> Extracts may be stored up to 7 days before analysis if storage is conducted under an inert (oxidant-free) atmosphere.
- $^{14}$  For the analysis of diphenylnitrosamine, add 0.008% Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub> and adjust pH to 7-10 with NaOH within 24 hours of sampling.
- <sup>15</sup> The pH adjustment may be performed upon receipt at the laboratory and may be omitted if the samples are extracted within 72 hours of collection. For the analysis of aldrin, add 0.008% Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub>.
- <sup>16</sup> Sufficient ice should be placed with the samples in the shipping container to ensure that ice is still present when samples arrive at the laboratory. However, even if ice is present when the samples arrive, it is necessary to immediately measure the temperature of the samples

and confirm that the  $6^{\circ}$ C temperature maximum has not been exceeded. In the isolated cases where it can be documented that this holding temperature can not be met, the permittee can be given the option of on-site testing or can request a variance. The request for a variance should include supportive data which show that the toxicity of the effluent samples is not reduced because of the increased holding temperature.

- <sup>17a</sup> Holding time is calculated from time of sample collection to elution for samples shipped to the laboratory in bulk and calculated from the time of sample filtration to elution for samples filtered in the field.
- <sup>17b</sup> Samples collected for the determination of trace level mercury (100 ng/L) using EPA Method 1631 must be collected in tightly-capped fluoropolymer or glass bottles and preserved with BrCl or HCl solution within 48 hours of sample collection. The time to preservation may be extended to 28 days if a sample is oxidized in the sample bottle. Samples collected for dissolved trace level mercury should be filtered in the laboratory. However, if circumstances prevent ovemight shipment, samples should be filtered in a designated clean area in the field in accordance with procedures given in Method 1669. Samples that have been collected for determination of total or dissolved trace level mercury must be analyzed within 90 days of sample collection.

SECTION 20. EFFECTIVE DATE. This rule shall take effect the first day of the month following publication in the Wisconsin administrative register as provided in s. 227.22(2)(intro.), Stats.

SECTION 21. BOARD ADOPTION. This rule was approved and adopted by the State of Wisconsin Natural Resources Board on August 11, 2004.

Dated at Madison, Wisconsin \_\_\_\_\_\_

STATE OF WISCONSIN DEPARTMENT OF NATURAL RESOURCES

Ву \_\_\_

Scott Hassett, Secretary

(SEAL)